

Fișă de lucru la tema „Proprietățile chimice ale apei”

Nr.	Ecuatia reacției	Tipul reacției
1	$\text{H}_2\text{O} + \text{SO}_2 \rightarrow$	
2	$\text{H}_2\text{O} + \dots\dots\dots \rightarrow \text{H}_2\text{SO}_4$	
3	$\text{H}_2\text{O} + \text{P}_2\text{O}_5 \rightarrow$	
4	$\text{Na} + \text{H}_2\text{O} \rightarrow$	
5	$\dots\dots + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2 + \text{H}_2$	
6	$\text{H}_2\text{O} + \text{Na}_2\text{O} \rightarrow$	
7	$\text{H}_2\text{O} + \dots\dots\dots \rightarrow \text{Ba}(\text{OH})_2$	
8	$\text{Mg} + \text{H}_2\text{O} \rightarrow$	
9	$\dots\dots + \text{H}_2\text{O} \rightarrow \text{ZnO} + \text{H}_2$	