

BOILING POINTS OF ORGANIC COMPOUNDS (PART 3: ALL ORGANIC COMPOUNDS)

DRAG THE ANSWERS TO THE CORRECT BOX

PREPARED BY NURUL HIDAYAH KAMARDIN



COMPOUNDS

1-butanol

1-butanamine

Ethanoic acid

pentanal

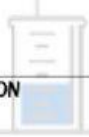


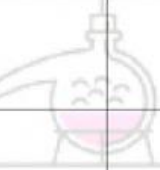
REASON

This compound can form weak van der Waals force

This compound can form stable hydrogen bonded dimers

This compound can form strong hydrogen bond because oxygen is more electronegative than nitrogen

This compound form weaker hydrogen bond because oxygen is more electronegative than nitrogen

	LOWEST BOILING POINT ----- HIGHEST BOILING POINT			
COMPOUND				
REASON				