

BOILING POINTS OF ORGANIC COMPOUNDS (PART 3: ALL ORGANIC COMPOUNDS)

DRAG THE ANSWERS TO THE CORRECT BOX

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COMPOUNDS

1-butanol

1-butanamine

Ethanoic acid

pentanal

REASON

This compound can form weak van der Waals force

This compound can form stable hydrogen bonded dimers

This compound can form strong hydrogen bond because oxygen is more electronegative than nitrogen

This compound form weaker hydrogen bond because oxygen is more electronegative than nitrogen

COMPOUND	LOWEST BOILING POINT				HIGHEST BOILING POINT	
REASON						
1-butanol						