


	Symbols & Oxidation #	Formula	Name of Compound
16. Sodium and phosphate	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>
17. Ammonium and hydroxide	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>
18. Calcium and nitrate	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>
19. Beryllium and hydroxide	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>
20. Potassium and acetate	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>
21. Sodium and carbonate	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>
22. Sodium and bicarbonate	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>
23. Aluminum and sulfate	<div><div></div><div><div></div><div></div></div><div><div></div><div></div></div></div>	<div><div></div><div></div></div> <div><div></div><div></div></div>	<div></div>

25. An ion is an atom or group of atoms that has become electrically

26. When an atom loses an electron its charge is (positive or negative)

27. An ionic bond is the attraction between (opposites, positive, neutral, or negative) ions.

28. Ionic compounds are electrically (charged, positive, neutral, or negative).

29. The sum of the charges for an ionic compound is 

The two answers must be in the right order.

30. An ionic compound is the result of the bonding of a (non-metal, metalloid, metal, noble gas) with a (non-metal, metalloid, metal).