

## Making Ionic Compounds

<b>Metal Ion (K<sup>+1</sup>, Ca <sup>+2</sup>)</b>	<b>How many metal ions in the compound?</b>	<b>Nonmetal ion (ex: Cl<sup>-1</sup>, F<sup>-1</sup>)</b>	<b>How many nonmetal ions in the compound?</b>	<b>Formula of the ionic compound</b>
Example:				