

NOMOR MASSA DAN NOMOR ATOM

NAMA :

KELAS :



Keterangan :

X = Lambang Unsur

A = Nomor Massa

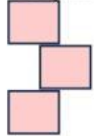
Z = Nomor Atom

Perlu diingat! Nomor Atom (Z) = Jumlah Proton (p) = Jumlah Elektron (e)

Jumlah Neutron (n) = Nomor Massa (A) – Nomor Atom (Z)

Latihan

1. Beryllium



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

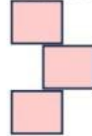
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

4. Carbon



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

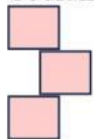
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

2. Sodium



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

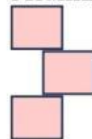
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

5. Aluminium



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

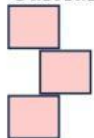
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

3. Calcium



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

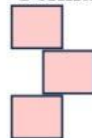
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

6. Germanium



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=