

Balancing Chemical Equations 2

1. $\text{Al(OH)}_3 + \text{HCl} \rightarrow \text{AlCl}_3 + \text{H}_2\text{O}$
2. $\text{CaCO}_3 + \text{HCl} \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$
3. $\text{Al} + \text{CuSO}_4 \rightarrow \text{Cu} + \text{Al}_2(\text{SO}_4)_3$
4. $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
5. $\text{MnO}_2 + \text{HCl} \rightarrow \text{MnCl}_2 + \text{H}_2\text{O} + \text{Cl}_2$
6. $\text{NaHCO}_3 + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O} + \text{CO}_2$
7. $\text{MgCO}_3 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O} + \text{CO}_2$
8. $\text{Mg(OH)}_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$
9. $\text{KNO}_3 \rightarrow \text{K}_2\text{O} + \text{N}_2 + \text{O}_2$
10. $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$