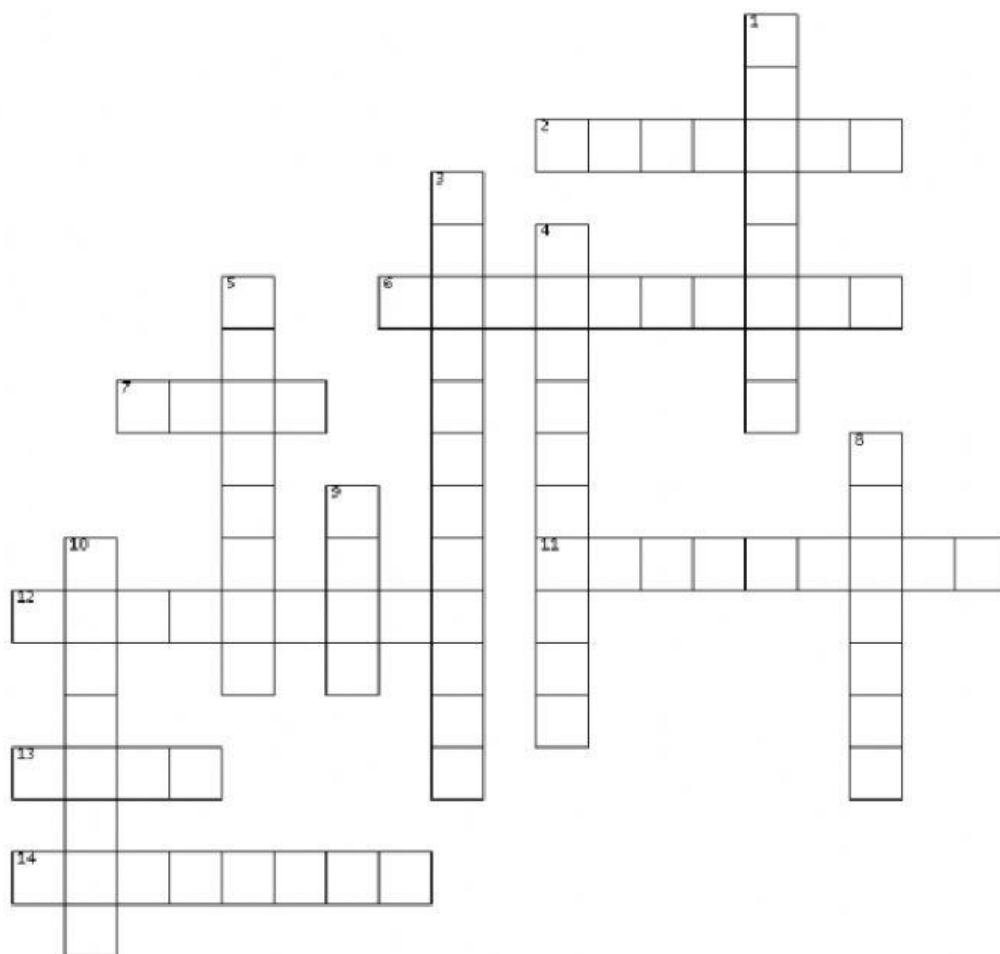


Chemical terms



ACROSS

- 2. (combination of pure substances not chemically combined)
- 6. (maximum amount of solute that will dissolve in a given amount of a solvent)
- 7. (mixture of a liquid with a gas)
- 11. (maximum amount of dissolved solute possible)
- 12. (substances composed of two or more elements fixed by a chemical bond)
- 13. (electroneutral particle)
- 14. (electroneutral particles consisting of two or more atoms fixed by a chemical bond)

DOWN

- 1. (written form of chemical reaction between reactants and products)
- 3. (studies chemical reactions and interactions that take place in living organisms)
- 4. (mixture of a solid in a liquid)
- 5. (process leading to chemical change in matter)
- 8. (positively charged particles)
- 9. (holds together atoms or ions in molecules or crystals)
- 10. (homogeneous mixture, composed of a solvent and one or more solutes)

boiling	atom	alloy
compound	brightness	bond
dilute	conductive	concentrated
element	electronegativity	ductility
liquid	homogeneous	equation
mixture	melting	malleability
reactant	particles	molecule
solid	saturated	reaction
solution	solute	solubility
	sublimation	solvent

D N U O P M O C A E H D D Q B O J Y I E
 X Y Y S S D S F V L I M M A L L O Y Z Q
 E F T T O U E I H U C O K X M F U R A U
 J U D I A L T T Q P T M O L E C U L E A
 X U U D V C V I A A A E D V U L Q M V T
 G A Z L U I L E E R T R H I Q E A D S I
 I E F D S V T Z N U T H T B L L U U G O
 U D N I O D Y A L T N N Q I L O B J K N
 E O H W L J U I G O Y N E E C L S J E Y
 C R N C U B D C I E S T A C I L A I L C
 N S U A T E R T T A N B I M N R E O E Y
 C U V T E C U E T I I O A L L O R S M M
 Y O S Y X L H U A L L T R Q I R C J E J
 Q I Z S O I R F I C I I K T E B B Y N A
 S G D S O A M T Q O T E T A C Y U S T O
 D I N R T S Y I N T P I C Y L E Y L T A
 G I M E B O I L I N G T O N H K L V O K
 N D D G N I T L E M A I T N U P U E X S
 R G M D E S Y B A N S S E N T H G I R B
 X Q B O N D H Z T H O M O G E N E O U S