



English – Class work



Look and Match. (Elige la opción correcta)

<input type="text"/>	CARBON 6 C 12.01	<input type="text"/>
<input type="text"/>		Usually from a Greek or Latin word for the element or a substance containing the element.
<input type="text"/>		Short-hand abbreviation for the element name.

The average mass of the atoms in an element.

Parts of an ATOM. (Arrastra los nombres a la definición correcta.)

Electron

Proton

Neutron

_____ Positively charged particle in the nucleus of the atom.

_____ The number of protons in an atom's nucleus is the atomic number.

_____ Negatively charged particle surrounding the nucleus of the atom.

_____ The number of electrons surrounding the nucleus of an atom is equal to the number of protons in the atom's nucleus.

_____ Particle in the nucleus that has about the same mass as a proton but has no charge. For the atoms of the first 20 elements, the number of neutrons is either equal to or slightly greater than the number of protons.