



English – Class work



Look and Match. (Elige la opción correcta)

<div data-bbox="172 398 466 474" style="border: 1px solid red; width: 184px; height: 34px; margin-bottom: 10px;"></div> <div data-bbox="197 499 512 584" style="text-align: center;">The number of protons in the nucleus of the atom.</div>	<div data-bbox="604 376 909 835" style="border: 1px solid black; padding: 10px; text-align: center;">CARBON 6 C 12.01</div>	<div data-bbox="1161 367 1476 443" style="border: 1px solid red; width: 197px; height: 34px; margin-bottom: 10px;"></div> <div data-bbox="1074 448 1407 629" style="text-align: center;">Usually from a Greek or Latin word for the element or a substance containing the element.</div> <div data-bbox="940 669 1244 739" style="border: 1px solid red; width: 191px; height: 31px; margin-top: 10px;"></div> <div data-bbox="944 754 1252 840" style="text-align: center;">Short-hand abbreviation for the element name.</div>
<div data-bbox="207 721 507 790" style="border: 1px solid red; width: 188px; height: 31px; margin-bottom: 10px;"></div> <div data-bbox="137 831 435 916" style="text-align: center;">The average mass of the atoms in an element.</div>		

Parts of an ATOM. (Arrastra los nombres a la definición correcta.)

Electron

Proton

Neutron

- _____ Positively charged particle in the nucleus of the atom.
The number of protons in an atom's nucleus is the atomic number.
- _____ Negatively charged particle surrounding the nucleus of the atom.
The number of electrons surrounding the nucleus of an atom is equal to the number of protons in the atom's nucleus.
- _____ Particle in the nucleus that has about the same mass as a proton but has no charge. For the atoms of the first 20 elements, the number of neutrons is either equal to or slightly greater than the number of protons.