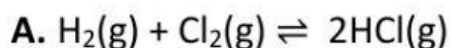


Name:

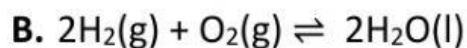
Class:

Starter: Equilibrium constant Expression

1. Write the equilibrium expressions for following chemical reactions at equilibrium



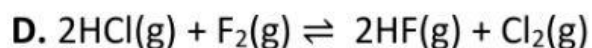
$K_{eq} = \dots\dots\dots$



$K_{eq} = \dots\dots\dots$



$K_{eq} = \dots\dots\dots$



$K_{eq} = \dots\dots\dots$

Recall:

$$K_{eq} = \frac{[\text{C}]^c [\text{D}]^d [\text{E}]^e}{[\text{A}]^a [\text{B}]^b}$$