Class 8 Chemistry

Calculate the molecular masses and write in the given space:

Use the following atomic masses:

N: 14, O: 16, P: 31, Na: 23, Al: 27, Si: 28, H: 1, Fe: 56, S: 32, Cl: 35.5, C: 12, K: 39, Ca: 40, Cu: 63.5

Molecular formula	Molecular mass	Molecular formula	Molecular mass
NO_2		K_2CO_3	
P_2O_5		$CaCO_3$	
Na_2O		$CuSO_4$	
Al_2O_3		$FeSO_4$	
SiO_2		$Ca(OH)_2$	
NH_3		HNO_3	
FeS		H_2SO_4	
H_2S		CH_3COOH	
$NaCl$		$C_{12}H_{22}O_{11}$	
$NaHCO_3$		$C_6H_{12}O_6$	