Nama: **Kelas:**

Jodohkanlah reaksi dibawah ini dengan jenis perubahan entalpi yang tepat !

$\Delta H^\circ f = 235$ Kj/mol	$KMnO_{4(s)} \rightarrow K_{(s)} + Mn_{(s)} + 2O_{2(g)}$
$\Delta H^\circ c = 127$ Kj/mol	$HCl_{(l)} + Mg(OH)_{2(l)} \rightarrow MgCl_{2(aq)} + 2H_2O_{(aq)}$
$\Delta H^\circ d = 321$ Kj/mol	$Ag_{(s)} + \frac{1}{2} Cl_{2(g)} \rightarrow AgCl_{(s)}$
$\Delta H^\circ n = 235$ Kj/mol	$H_2SO_{4(l)} \rightarrow H_2SO_{4(aq)}$
$\Delta H^\circ sol = 235$ Kj/mol	$S_{(s)} + O_{2(g)} \rightarrow SO_{2(g)}$

Ayuk semangat !!!

LIVEWORKSHEETS