

## REVIEW CHAPTER 6

1. Which of these measures would reduce the quantity of acid rain produced?

- I. Removing SO<sub>2</sub> from smokestacks
  - II. Switching to renewable energy
  - III. Using lime to neutralize acidified lakes
  - IV. Replacing coal with natural gas as energy source
- A. I, II and IV only
  - B. II and IV only
  - C. I, II and III only
  - D. I, III and IV only

(Total 1 mark)

2. Which is most likely to occur as a result of an **increase** in stratospheric ozone?

- A. Increased damage to plant life
- B. Reduced concentration of CFCs in the stratosphere
- C. Increased formation of photochemical smog
- D. Reduction in ultraviolet radiation reaching the Earth's surface

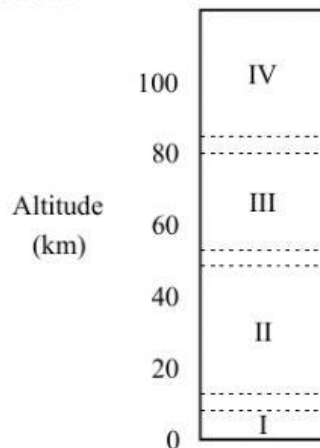
(Total 1 mark)

3. Photochemical smog is part of a cyclic process which requires \_\_\_I\_\_\_ and produces \_\_\_II\_\_\_.  
Select the words that should be inserted from the table below.

|    | I                                        | II                    |
|----|------------------------------------------|-----------------------|
| A. | ultraviolet radiation                    | CFCs                  |
| B. | sun light and nitrogen oxides            | oxygen atoms          |
| C. | sun light and water vapour               | ultraviolet radiation |
| D. | ultraviolet radiation and carbon dioxide | carbon monoxide       |

(Total 1 mark)

4. This question refers to the graph below.



The troposphere is

- A. I.
- B. II.
- C. III.
- D. IV.

(Total 1 mark)

5. Lime was put into Scandinavian lakes in the 1980s as

- A. a response to acid rain from the combustion of fossil fuels in other European countries.
- B. an attempt to replace nitrogen lost by excessive harvesting of fish.
- C. an attempt to combat eutrophication caused by excessive local use of fertilisers.
- D. a waste product from local industrial processes.

(Total 1 mark)

6. Which row contains correct information on the causes and effects of acid rain?

|    | Caused by                         | Effect                                                         |
|----|-----------------------------------|----------------------------------------------------------------|
| A. | Halogenated gases and nitric acid | Decrease in oceanic uptake of CO <sub>2</sub> by phytoplankton |
| B. | Sulfuric acid and methane         | Erosion of buildings and monuments                             |
| C. | Carbon dioxide and sulfur dioxide | Leaching of calcium ions from soil                             |
| D. | Carbon dioxide and nitric acid    | Increase in skin cancer                                        |

(Total 1 mark)

7. The ozone layer can be considered as natural capital which is

- A. replenishable and provides an essential service.
- B. replenishable and provides valuable goods.
- C. renewable and provides an essential service.
- D. renewable and provides valuable goods.

(Total 1 mark)

8. Which is the most likely to occur as a result of a reduction of stratospheric ozone?

- A. Decrease in the ultraviolet radiation reaching the Earth's surface
- B. Reduction of photochemical smog
- C. Reduction in the productivity of phytoplankton
- D. Increased leaching of aluminium ions attached to soil minerals

(Total 1 mark)

9. Problems with replacing CFCs (chlorofluorocarbons) with HFCs (hydrofluorocarbons) are that HFCs

- I. are powerful greenhouse gases.
  - II. remain in the atmosphere much longer than CFCs.
  - III. are expensive to produce.
- A. I and II only
  - B. II and III only
  - C. I and III only
  - D. I, II and III

(Total 1 mark)

10. Which of these statements about ultra-violet (UV) radiation are true?

- I. It is absorbed by most greenhouse gases.
  - II. It reduces productivity in phytoplankton.
  - III. It is harmful long-wave radiation coming from the sun.
- A. I, II and III
  - B. I and III only
  - C. II and III only
  - D. II only

(Total 1 mark)

11. Which processes occur in the following regions of the atmosphere?

|    | Troposphere                                 | Lower Stratosphere                          | Upper Stratosphere                          |
|----|---------------------------------------------|---------------------------------------------|---------------------------------------------|
| A. | Natural formation of ozone                  | Destruction of ozone through human activity | Formation of ozone through human activity   |
| B. | Formation of ozone through human activity   | Destruction of ozone through human activity | Natural formation of ozone                  |
| C. | Destruction of ozone through human activity | Natural formation of ozone                  | Formation of ozone through human activity   |
| D. | Formation of ozone through human activity   | Natural formation of ozone                  | Destruction of ozone through human activity |

12. Which impact(s) on ecosystems may be associated with the release of sulfur oxides into the atmosphere?
- Increased uptake of aluminium ions by living organisms
  - A decrease in certain mineral storages in the soil
  - Reduced leaf surface area in coniferous forests
- A. I, II and III  
 B. I and III only  
 C. I only  
 D. III only

(Total 1 mark)

13. Which column in the table correctly shows the effects of the pollutant gas?

|                                 | A.                     | B.                       | C.             | D.                    |
|---------------------------------|------------------------|--------------------------|----------------|-----------------------|
|                                 | <b>Sulphur dioxide</b> | <b>Halogenated gases</b> | <b>Methane</b> | <b>Carbon dioxide</b> |
| increases the greenhouse effect | Yes                    | No                       | Yes            | Yes                   |
| depletes stratospheric ozone    | Yes                    | Yes                      | No             | Yes                   |
| increases acidity of rain       | Yes                    | No                       | No             | Yes                   |

(Total 1 mark)

14. The ozone layer can be protected by
- using substitutes for ozone-depleting chemicals.
  - reducing cattle production.
  - recycling old refrigerators.
- A. I and II only  
 B. I and III only  
 C. II and III only  
 D. I, II and III

(Total 1 mark)

15. The Montreal Protocol was an agreement on
- slowing down world population growth.
  - reducing carbon dioxide emissions.
  - reducing the production of chlorofluorocarbons.
  - reducing the number of coal-fired power stations.

(Total 1 mark)

16. Which is the correct set of statements?

|    | <b>Troposphere</b>                                                                                | <b>Stratosphere</b>                                                                               |
|----|---------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------------------------------|
| A. | Absorbs large proportions of long-wave radiation from above; temperature increases with altitude  | Absorbs large proportions of short-wave radiation from below; temperature decreases with altitude |
| B. | Absorbs large proportions of long-wave radiation from below; temperature decreases with altitude  | Absorbs large proportions of short-wave radiation from above; temperature increases with altitude |
| C. | Absorbs large proportions of short-wave radiation from above; temperature increases with altitude | Absorbs large proportions of long-wave radiation from below; temperature decreases with altitude  |
| D. | Absorbs large proportions of short-wave radiation from below; temperature decreases with altitude | Absorbs large proportions of long-wave radiation from above; temperature increases with altitude  |

17. In a country in which most energy is obtained from burning fossil fuels, which of the following will tend to reduce the amount of acid deposition?
- Switching from conventional to nuclear power stations
  - Reducing tax on fossil fuels
  - Using lead-free petrol (gasoline) in cars
  - Banning the use of substances that harm the ozone layer, such as CFCs

(Total 1 mark)

18. Which row contains correct statements about ozone depletion and global warming?

|    | Ozone depletion is ...                         | Greenhouse effect is ...                         |
|----|------------------------------------------------|--------------------------------------------------|
| A. | a necessary condition for life on Earth.       | caused by ultraviolet radiation.                 |
| B. | caused by ultraviolet radiation.               | caused by carbon dioxide, methane and CFCs.      |
| C. | a cause of damage to photosynthetic organisms. | a cause of increases in the rate of skin cancer. |
| D. | caused by CFCs.                                | a necessary condition for life on Earth.         |

(Total 1 mark)

19. Why is lime sometimes added to acid lakes?

- |                                 |                              |
|---------------------------------|------------------------------|
| A. To lower the pH of the water | C. To kill poisonous fish    |
| B. To raise the pH of the water | D. To prevent eutrophication |

(Total 1 mark)

20. Which statement about the ozone layer is correct?

- Fluorine released from chlorofluorocarbons combines with carbon dioxide molecules in the upper atmosphere to form ozone.
- Acid rain is caused by an increase in the amount of ultraviolet radiation that penetrates the ozone layer.
- The burning of fossil fuels causes the formation of ozone, which gradually diffuses to the stratosphere.
- Ultraviolet radiation is absorbed during the process of stratospheric ozone formation.

(Total 1 mark)

21. Which is the most likely to occur as a result of a reduction of stratospheric ozone?

- Increase in the amount of damage to plant life at the Earth's surface
- Reduction of photochemical smog
- Reduction in ultraviolet radiation reaching the Earth's surface
- Increase in human respiratory problems

(Total 1 mark)

22. Which of the following is **not** a result of acid deposition from burning of fossil fuels?

- Leaching of calcium from soils
- Death of coniferous trees in forests
- Killing of fish due to high levels of aluminium in lakes
- Thermal expansion of oceans

(Total 1 mark)

23. Why is powdered limestone sometimes spread on lakes?

- To encourage the growth of a sustainable crop of fish
- To decrease the alkalinity of the water
- To reduce the acidity caused by air pollution
- To neutralize the effect of methane escaping from organic matter

(Total 1 mark)

24. What is a reduction in the amount of ozone in the upper atmosphere most likely to lead to?

- |                                                     |                                 |
|-----------------------------------------------------|---------------------------------|
| A. Damage to plant tissues by ultraviolet radiation | C. A decrease in rainfall       |
| B. Rapid global warming                             | D. An increase in acid rainfall |

(Total 1 mark)

25. What is the possible result of depletion of stratospheric ozone?

- Increased acidification of rain
- Reduced amounts of energy fixed by phytoplankton
- Increased levels of greenhouse gases
- Reduced levels of ultraviolet light reaching the Earth's surface

26. Which of these human activities both increases global warming and depletes the ozone layer?
- Emission of carbon dioxide from vehicle exhausts
  - Emission of sulfur dioxide from power stations
  - Leakage of methane from gas pipelines
  - Release of CFCs from old refrigerators

(Total 1 mark)

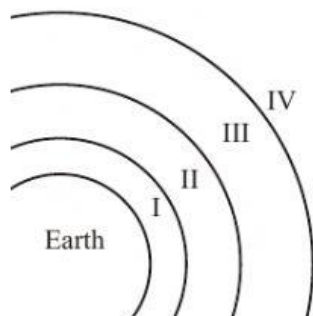
27. Which method would be most effective in reducing the acidity of rain downwind of a coal-fired power station?
- Burning high sulfur coal
  - Increasing the temperature at which fuel is burned
  - Fitting wet scrubbers (filters) to chimneys
  - Converting the power station to burn high sulfur oil

(Total 1 mark)

28. Which factor greatly increases the likelihood of photochemical smog forming over a city?
- Low altitude
  - High rainfall
  - High traffic densities
  - Short daylight hours

(Total 1 mark)

29. The figure below represents the different layers of the atmosphere



Which layers correspond to the troposphere and the stratosphere?

|    | Troposphere | Stratosphere |
|----|-------------|--------------|
| A. | I           | II           |
| B. | I           | III          |
| C. | II          | III          |
| D. | II          | IV           |

(Total 1 mark)

30. Which statement is correct?
- Ozone is destroyed by carbon dioxide released by burning of fossil fuels.
  - During recent decades the ozone hole over the Antarctic has become larger, producing global warming.
  - Ultraviolet radiation is involved in the production of stratospheric ozone.
  - Ozone is essential for human respiration.

(Total 1 mark)

31. Which of the following are possible effects of acid deposition?
- Death of fish due to increasing levels of aluminium in lakes
  - Reduction of forest productivity due to depletion of soil nutrients
  - Death of trees due to decreasing levels of aluminium in soils
- I and II only
  - II and III only
  - I and III only
  - I, II and III

(Total 1 mark)

32. Which statement about photochemical smog is correct?
- A. Photochemical smog is independent of local topography and climate.
  - B. Photochemical smog is a mixture of snow and rain formed when the upper air temperature is below freezing.
  - C. Photochemical smog results from the meeting of two air streams in areas of low pressure.
  - D. Photochemical smog is a mixture of pollutants formed under the influence of sunlight.

(Total 1 mark)

33. Which atmospheric gas filters out harmful ultraviolet radiation?
- A. Oxygen
  - B. Ozone
  - C. Carbon dioxide
  - D. CFCs

(Total 1 mark)

34. Which of the following are normally associated with the formation of photochemical smog?
- I. Nitrogen oxides
  - II. Sunlight
  - III. Volatile organic compounds
  - IV. Clouds
- A. I and III only
  - B. III and IV only
  - C. I, II and III only
  - D. I, II, III and IV

(Total 1 mark)

35. For which environmental impacts has acid deposition been identified as a significant cause?
- I. Leaching of calcium from soil
  - II. Increasing the rate of respiratory problems in humans
  - III. Contamination of fish with toxic aluminium
  - IV. Increasing the productivity of marine phytoplankton
- A. I and II only
  - B. I and IV only
  - C. I, II and III only
  - D. I, II, III and IV

(Total 1 mark)

36. Which of the following is most likely to lead to increased mutation rates in phytoplankton?
- A. Increase in stratospheric ozone
  - B. Decrease in stratospheric ozone
  - C. Increase in tropospheric ozone
  - D. Decrease in tropospheric ozone

(Total 1 mark)

37. Which of the following has been most effective in reducing world emissions of CFCs?
- A. Improved technologies in the development of electric cars
  - B. Aid programmes to developing countries
  - C. Policies to minimize the combustion of fossil fuels
  - D. International agreements, including the Montreal Protocol

38. What happens when sunlight interacts with pollutants in the lower atmosphere?
- A. Destruction of the ozone layer occurs
  - B. Stratospheric ozone is produced
  - C. Thermal inversion occurs
  - D. Photochemical smog is produced

(Total 1 mark)

40. Which statement about acid rain is correct?

- A. Acid rain is a problem of developed countries because it is always deposited onto industrial areas.
- B. Rain with pH above 7 is considered to be acid rain.
- C. Lime can be used to restore acidified lakes.
- D. Corrosion of buildings and rising sea levels are two of the main effects of acid rain.

(Total 1 mark)

41. Which of the following pairs of statements about ozone is correct?

|    | <b>Tropospheric ozone</b>                                    | <b>Stratospheric ozone</b>                                   |
|----|--------------------------------------------------------------|--------------------------------------------------------------|
| A. | It is significantly affected by increasing levels of CFCs    | It accumulates in high concentration to form the ozone layer |
| B. | It causes health problems in humans and animals              | It is significantly affected by increasing levels of CFCs    |
| C. | It accumulates in high concentration to form the ozone layer | It damages photosynthetic organisms                          |
| D. | It damages photosynthetic organisms                          | It causes health problems in humans and animals              |

(Total 1 mark)