Le Chatelier's Principle

For this assignment use the following (**ALL LOWERCASE**):

r for **r**ight l for **l**eft i for **i**ncrease d for **d**ecrease s for **s**ame	Is the equation below exothermic or endothermic? Spell it out, lowercase, next to the equation. When you are done click finish, it will give you a grade. Upload your screenshot to the assignment.

$$12.6\ kcal + H_2(g) + I_2(g) \leftrightarrow 2HI(g)$$

Stress	Equilibrium Shift	$[H_2]$	$[I_2]$	[HI]	K
1. Add H_2	right	——	decreases	increases	remains the same
2. Add I_2			——		
3. Add HI				——	
4. Remove H_2		——			
5. Remove I_2			——		
6. Remove HI				——	
7. Increase Temperature					
8. Decrease Temperature					
9. Increase Pressure					
10. Decrease Pressure					

LIVEWORKSHEETS