

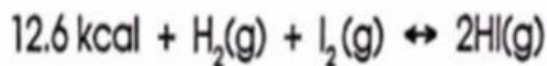
# Le Chatelier's Principle

For this assignment use the following (ALL LOWERCASE):

r for right  
l for left  
i for increase  
d for decrease  
s for same

Is the equation below exothermic or endothermic?  
Spell it out, lowercase, next to the equation.

When you are done click finish, it will give you a grade. Upload your screenshot to the assignment.



Stress	Equilibrium Shift	[H <sub>2</sub> ]	[I <sub>2</sub> ]	[HI]	K
1. Add H <sub>2</sub>	right	_____	decreases	increases	remains the same
2. Add I <sub>2</sub>			_____		
3. Add HI				_____	
4. Remove H <sub>2</sub>		_____			
5. Remove I <sub>2</sub>			_____		
6. Remove HI				_____	
7. Increase Temperature					
8. Decrease Temperature					
9. Increase Pressure					
10. Decrease Pressure					