

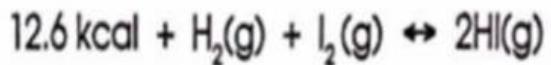
## Le Chatelier's Principle

For this assignment use the following (ALL LOWERCASE):

r for right  
l for left  
i for increase  
d for decrease  
s for same

Is the equation below exothermic or endothermic?  
Spell it out, lowercase, next to the equation.

When you are done click finish, it will give you a grade. Upload your screenshot to the assignment.



Stress	Equilibrium Shift	$[\text{H}_2]$	$[\text{I}_2]$	$[\text{HI}]$	$K$
1. Add $\text{H}_2$	right	—	decreases	increases	remains the same
2. Add $\text{I}_2$			—		
3. Add $\text{HI}$				—	
4. Remove $\text{H}_2$		—			
5. Remove $\text{I}_2$			—		
6. Remove $\text{HI}$				—	
7. Increase Temperature					
8. Decrease Temperature					
9. Increase Pressure					
10. Decrease Pressure					