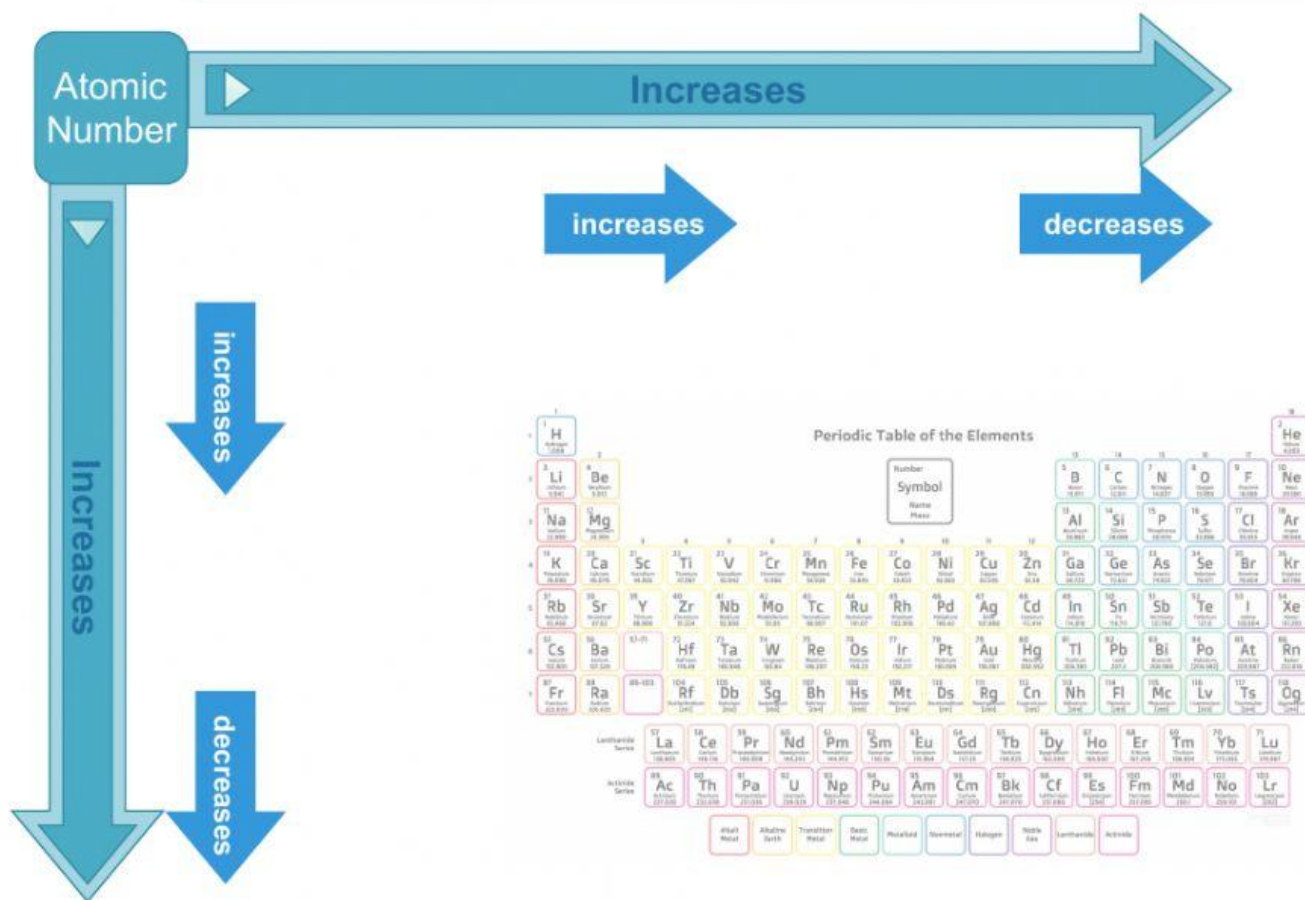


Graduation of elements properties in modern periodic table



- By increasing atomic number in groupincrease due to increasing number of energy levels.
- By increasing atomic number in period atomic size due to increasing attraction force between nucleus and outer energy level's electrons.
- electronegativity increases by increasing atomic number inand decreases by increasing atomic number in
- strongest metals found atside of modern periodic table, while strongest non metals found at theside before inert gas
- the metalloid is found betweenelement andelement
- The measuring unit of atomic radius is
- metal oxide isoxide, while non-metal oxide isoxide