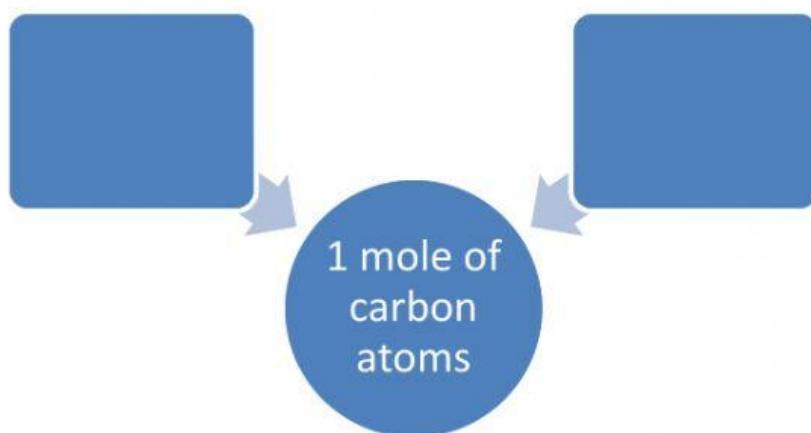
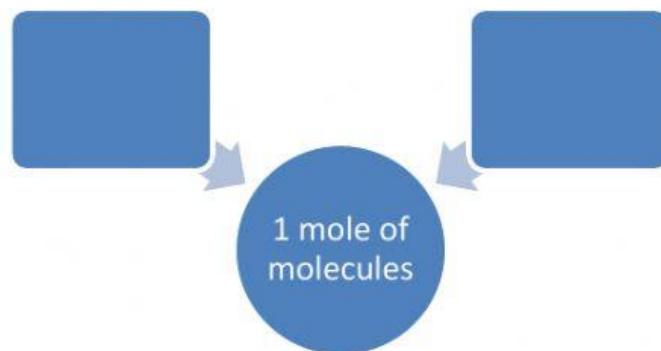
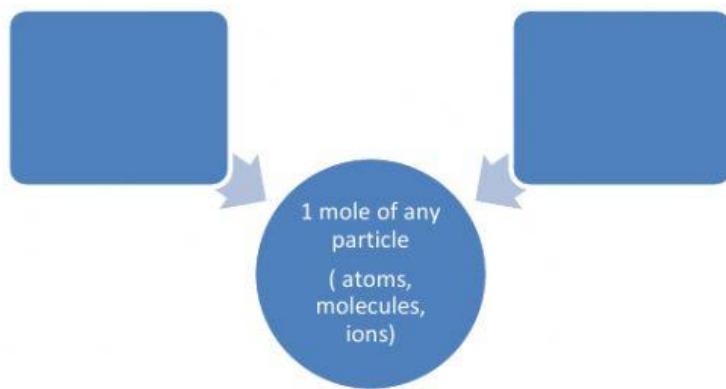


## GRADE 9 MOLE CONCEPT



$6.022 \times 10^{23}$   
atoms of  
Carbon

$6.022 \times 10^{23}$   
atoms of  
hydrogen

$6.022 \times 10^{23}$   
number of  
that particle

$6.022 \times 10^{23}$   
numbers of  
molecule

12 g of  
carbon atom

Relative mass of  
those particles in  
grams

Molecular mass in  
grams