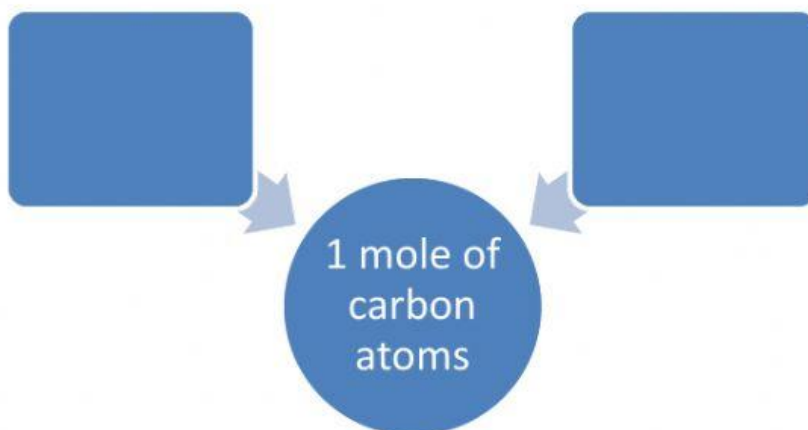
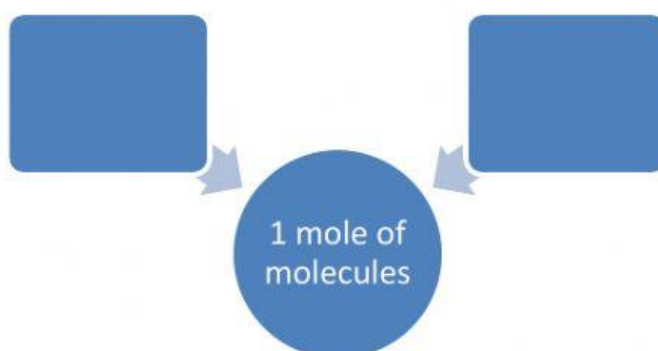
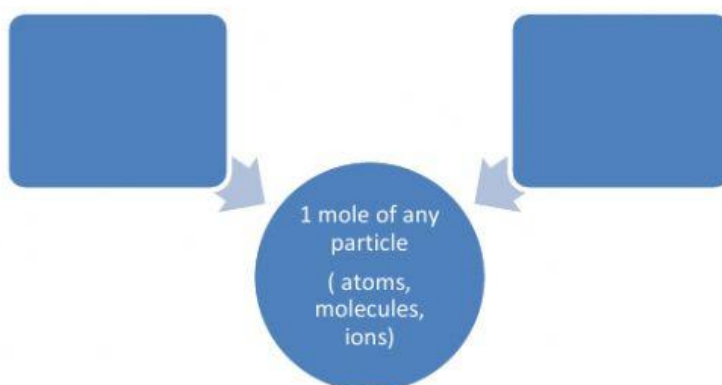


## GRADE 9 MOLE CONCEPT



$6.022 \times 10^{23}$ atoms of Carbon	$6.022 \times 10^{23}$ atoms of hydrogen	$6.022 \times 10^{23}$ number of that particle	$6.022 \times 10^{23}$ numbers of molecule
12 g of carbon atom	Relative mass of those particles in grams	Molecular mass in grams	