Lesson: Natural Science.
Topic: Atoms

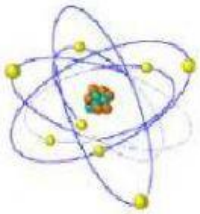
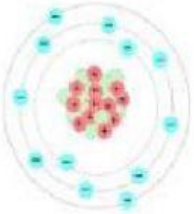
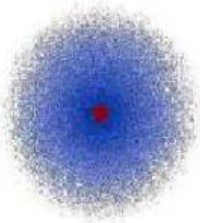
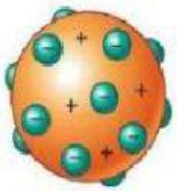
1)Complete the sentence.

Mass **space** **place** **atoms** **broken**

- Matter: Is everything that surrounds you, has _____ and occupies a _____ in _____.

- Substances that are formed by the same _____, that cannot be _____into anything simpler.

2) Relate the atomic model to the scientist who discovered it.

			Rutherford	
Schrödinger	Thomson	~~Rutherford~~	Dalton	Bohr

3) Line the names with their definition.

Enlace químico	Es la unión de dos o más átomos, debido a la fuerza de atracción o repulsión de los electrones.
Fórmula química	Muestran los elementos que forman el compuesto y la proporción que existe entre ellos. Example: H2O.

LIVEWORKSHEETS

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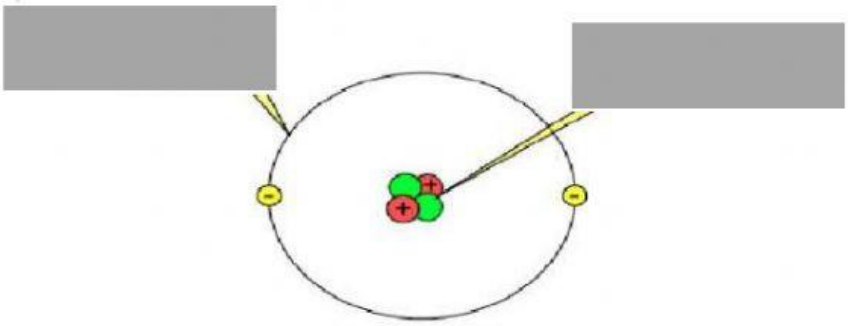
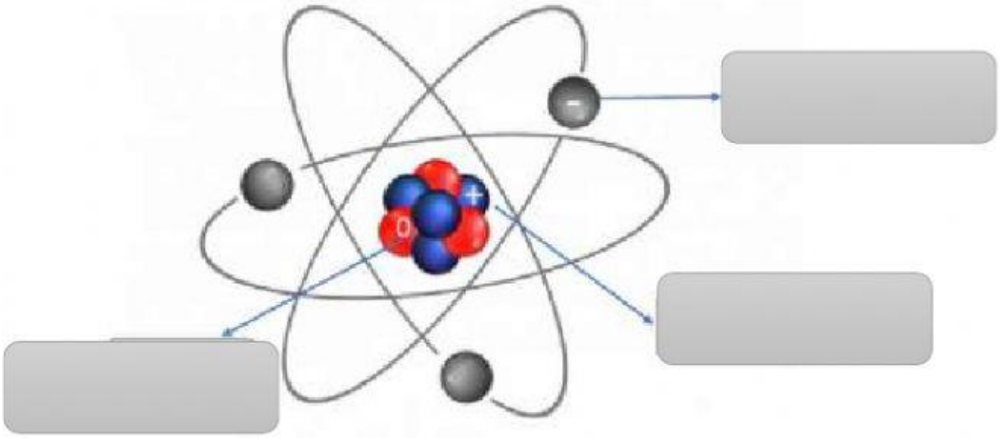
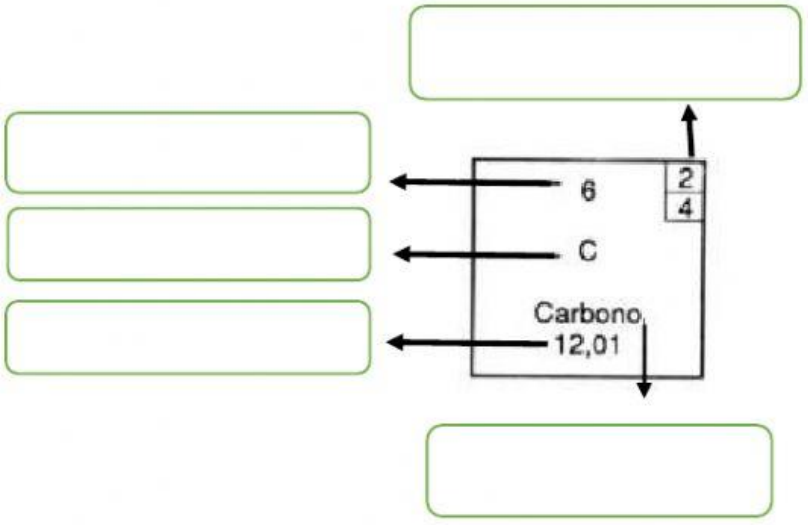
4) Fill in the blanks with the names of the parts of the atom.



5) Write the name of the subatomic particles where applicable.



6)Identify what each part of the image refers to.



Distribución de electrones en cada nivel.

Nombre

Número atómico

Símbolo

Masa atómica

LIVEWORKSHEETS