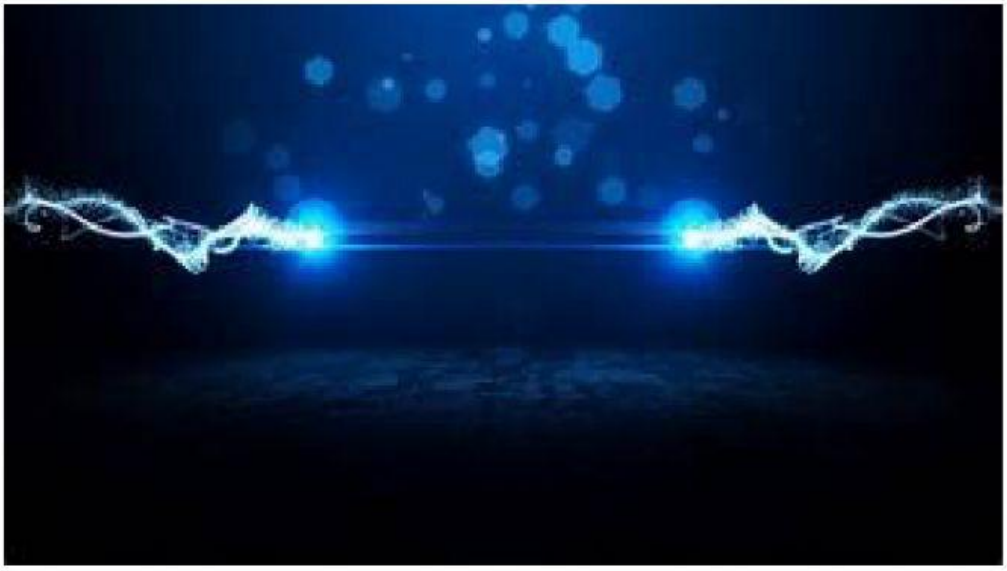
Watch the video before you proceeed with the Exercise 4



Proceed to the Exercie 4 below

Exercise 4: Complete the table below with proper formula and name of the ionic compounds

	Cation	Anion	Formula	Name
1.	NH_4^+	Cl^-		
2.	Fe^{3+}	NO_3^-		
3.	Ti^{3+}	Br^+		
4.	Cu^+	P^{3-}		
5.	Pb^{2+}	SO_4^{2+}		
6.	Al^{3+}	CN^-		
7.	Sn^{2+}	NO_2^-		
8.	NH_4^+	Cl^-		
9.	Ca^{2+}	Cl^-		
10.	Li^+	I^-		
11.	Pb^{2+}	N^{3-}		
12.	Ag^+	Br^-		
13.	Na^+	F^-		
14.	NH_4^+	Cl^-		
15.	Ca^{2+}	CO_3^{2-}		
16.	Zn^+	P^{3-}		
17.	Li^+	SO_3^{2-}		
18.	Cu^+	O^{2-}		
19.	K^+	MnO_4^-		
20.	Fe^{3+}	Cl^-		