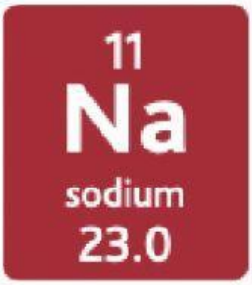
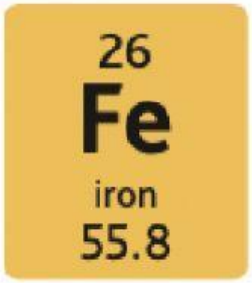
In the space below, write the <u>unabbreviated</u> electron configurations of the following elements:

Drag the correct electron configuration to the correct element below.





$1s^2 2s^2 2p^6 3s^1$

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$

$1s^2 2s^2 2p^4$

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Drag the correct element to the correct electron configuration

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$

$1s^2 2s^2 2p^6 3s^2 3p^4$

$[Kr]\ 5s^2 4d^{10} 5p^3$

16
S
sulfur
32.1

37
Rb
rubidium
85.5

51
Sb
antimony
121.8

LIVEWORKSHEETS