

Complete the following sentence:

Isotopes are atoms of same element with the same but different _____

Look at the isotope notation below:

$$^{200}_{80}\,Hg^{2+}$$

Choose the correct number of sub-atomic particles:

Proton Neutron Electron

Choose the correct statement about carbon

Component of graphite

Consist of only one isotope

It is positively charge

Which of the following is correct about neutron?

It is neutral
It is in the nucleus
It is positively charge





Match the following subatomic particle and its charge

Proton	Negative
	11090000

Neutron Positive

Electron Neutral

Drag the answers to the correct place

Negative Positive Neutral

Subatomic particle	Charge
Proton	
Neutron	
Electron	



