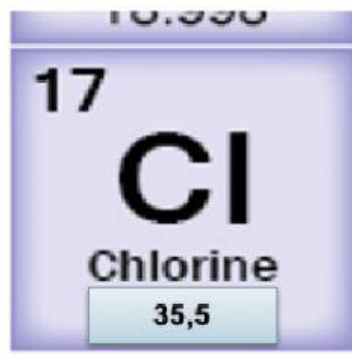


Practice

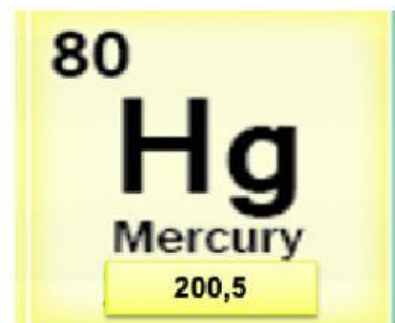
-Complete the PEN of each element below.



Proton :
Electron :
Neutron :



Proton :
Electron :
Neutron :



Proton :
Electron :
Neutron :

Summarize

1. What is the atomic mass of the element shown?

a.) 18 b.) 13 c.) 12 d.) 11

2.) How many electrons would there be in a neutral atom of the element above?

a.) 6 b.) 12 c.) 6.011 d.) 18

6
C
Carbon
12.011

3.) T/F: The periodic table is arranged by atomic mass

4.) T/F: Most of the mass of an element is in the electron shell.