Drug una a	rop, some m	oras may not	De asea.		
temperatur			inwards	mass ze	ro
	decreases	decreases			
increases		increases	increases	increases	
weight	stops				
1. Boyle's law	- At constant	1.	as the p	ressure incr	reases
the volum	1 <b>e</b> 2.	. As the p	ressure dec	reases the v	olume 3.
Pressure	up Volume 4.	Charles Administrations of the Commission			
	Section - Committee of the Committee of				
Pressure (	down Volume	3 5.			
2 Charles's las	at consta	<b>*</b> 6	ac t	no tomporati	uro inorocco
2. Charles's law - at constant 6. as the temperature increases the volume 7. As the temperature decreases the volume 8.					
the volum	e "	. AS L	ne temperati	ire decrease	is the volume 8.
3. pressure - in	a container	t is caused b	y the particle	s bumping i	nto the sides of
the contain	ner and push	ing <sup>9.</sup>			
In the atmo	osphere it is	the 10.	of the air	above us.	
		ophere down		,	
	- p	opiloro do un			
4. absolute zer	o -This is the	lowest poss	ible temperat	ure, the mot	ion of
the particle	es 11.	and the vo	lume goes to	12.	
Secretaria de Caracteria de Ca					
5. If the temper	ature remains	constant, what	will happen to	the pressure of	of a gas if you decrease
the volume o	f the container	that holds it?	13.		
					MILE SECTIONS OF A CONTRACT AND A SECTION OF A CONTRACT AND A CONT
			s constant, wha	it will happen	to the pressure of a gas
if you increa	se temperature	? 14.			
Directions: An	swer the followin	ig questions regai	raing temperaturi	e. Fili in you	r own answer
7. On the Kel	vin scale, what	is the freezing	point of water?		
8. On the Kel	vin scale, what	is the boiling p	ooint of water?		
0 On the Cal	alian anala suka	the fore-!-	a and ballings	sainte of weets	-2 Freezing
9. On the Cel	sius scare, wha	are the freezin	ig and boiling p	ounts of water	r? Freezing Boiling
					DOMING

**#LIVEWORKSHEETS**