

Nama :

Kelas :

BILANGAN KUANTUM

Lengkapilah tabel berikut ini!

Unsur	Konfigurasi elektron	Elektron terakhir			
		n	l	m	s
${}_5\text{B}$	$1s^2 2s^2 2p^1$				
${}_{11}\text{Na}$	$1s^2 2s^2 2p^6 3s^1$				
${}_{15}\text{P}$	$1s^2 2s^2 2p^6 3s^2 3p^3$				
${}_{24}\text{Cr}$	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^4$				
${}_{27}\text{Co}$	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^7$				
${}_{30}\text{Zn}$	$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10}$				
${}_{10}\text{Ne}$					
${}_{18}\text{Ar}$					
${}_{23}\text{V}$					
${}_{14}\text{Si}$					
${}_{20}\text{Ca}$					
${}_9\text{F}$					

