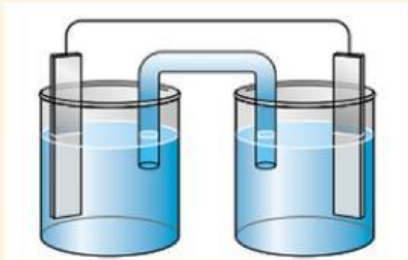


# LKPD

NAMA : .....

KELAS : .....

## MARI BERLATIH SOAL SEL VOLTA



The picture on the side is a picture of a voltaic cell / galvanic cell having lead (Pb) and zinc (Zn) electrodes.

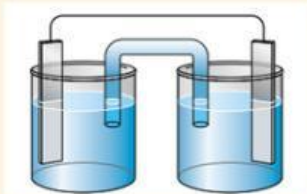


### Pertanyaan :

- 1 Label the electrodes, identify the ions in the solution, and label the anode and cathode!
- 2 Indicate the direction of electron flow and the direction of ions in solution



## MARI BERLATIH SOAL SEL VOLTA



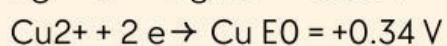
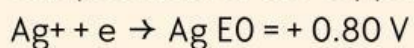
**A.** In a Voltaic cell, the iron anode is immersed in  $\text{FeSO}_4$ , the nickel cathode is immersed in  $\text{NiSO}_4$ . Make the cell reaction and cell notation!



### Jawaban



The potential of the copper silver electrode is as follows:



Find:

- Cell diagram
- Cell potential price Cell reaction
- Cell reaction

