NOMOR MASSA DAN NOMOR ATOM

NAMA :

KELAS :

$${}^{A}_{Z}X$$

Keterangan :

X = Lambang Unsur

A = Nomor Massa

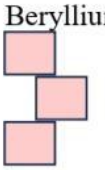
Z = Nomor Atom

Perlu diingat! Nomor Atom (Z) = Jumlah Proton (p) = Jumlah Elektron (e)

Jumlah Neutron (n) = Nomor Massa (A) – Nomor Atom (Z)

Latihan

1. Beryllium



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

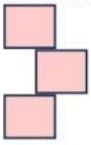
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

2. Sodium



Nomor Massa (A) =

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Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

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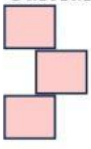
3. Calcium



Nomor Massa (A) =

Nomor Atom (Z) =

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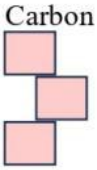
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

4. Carbon



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

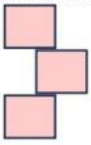
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

5. Aluminium



Nomor Massa (A) =

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Jumlah Proton (p) =

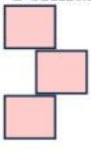
Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

= -

=

6. Germanium



Nomor Massa (A) =

Nomor Atom (Z) =

Jumlah Proton (p) =

Jumlah Elektron (e) =

Jumlah Neutron (n) = A – Z

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