

1. $\text{Zn (s)} + \text{CuSO}_4 \text{ (aq)} \rightarrow \text{Cu (s)} + \text{ZnSO}_4 \text{ (aq)}$
2. $\text{Ba (NO}_3)_2 \text{ (aq)} + \text{Na}_2\text{SO}_4 \text{ (aq)} \rightarrow \text{BaSO}_4 \text{ (s)} + \text{NaNO}_3 \text{ (aq)}$
3. $\text{CoCl}_2 \text{ (aq)} + \text{Na (s)} \rightarrow \text{NaCl (aq)} + \text{Co (s)}$
4. $\text{Zn (s)} + \text{HCl (aq)} \rightarrow \text{ZnCl}_2 \text{ (aq)} + \text{H}_2 \text{ (g)}$
5. $\text{BaCl}_2 \text{ (aq)} + \text{CuSO}_4 \text{ (aq)} \rightarrow \text{BaSO}_4 \text{ (s)} + \text{CuCl}_2 \text{ (aq)}$
6. $\text{AgNO}_3 \text{ (aq)} + \text{Cu (s)} \rightarrow \text{CuNO}_3 \text{ (aq)} + \text{Ag (s)}$
7. $\text{NaNO}_3 \text{ (aq)} + \text{K (s)} \rightarrow \text{KNO}_3 \text{ (aq)} + \text{Na (s)}$