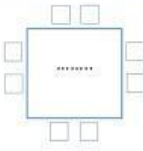
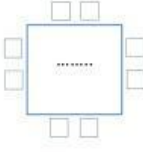
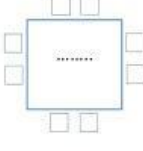
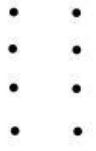
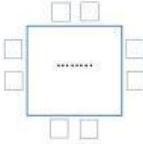
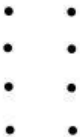
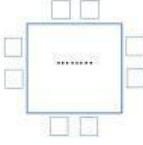
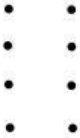


APLIKASI

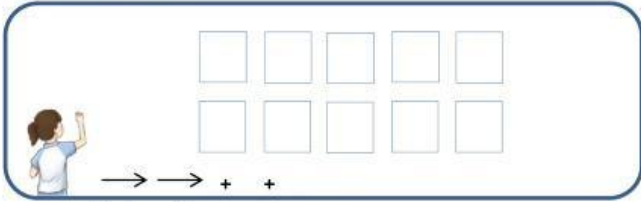
1. Lengkapi tabel dibawah ini!

Unsur	Konfigurasi Elektron	Elektron Valensi	Melepas/Menerima Elektron	Konfigurasi Elektron Baru	Lambang Ion	Struktur Lewis	
${}_4\text{Be}$							
${}_{17}\text{Cl}$							
${}_3\text{Li}$							

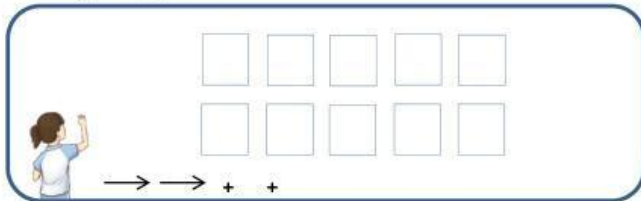


^{16}S								
^{20}Ca								
^{35}Br								

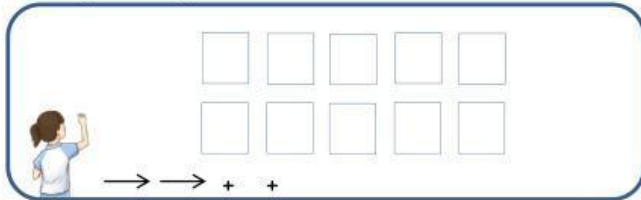
2. Gambarkan proses terjadinya ikatan ion antara unsur-unsur berikut!!
- a) Tuliskan reaksi pembentukan ion positif dan ion negatif dari unsur-unsur berikut
- ${}_8\text{Be}$ dan ${}_{16}\text{S}$



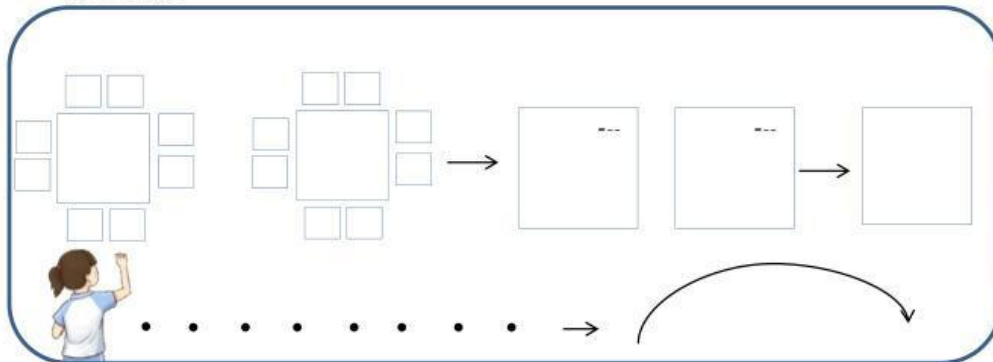
- ${}_3\text{Li}$ dan ${}_{17}\text{Cl}$



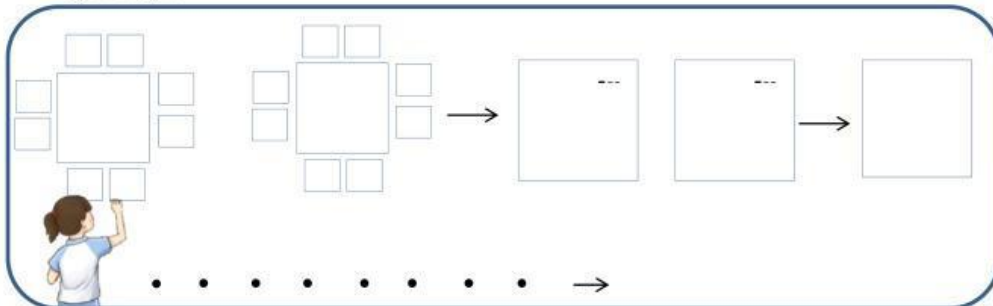
- ${}_{20}\text{Ca}$ dan ${}_{35}\text{Br}$



- b) Tuliskan skema perpindahan elektron menggunakan struktur lewis seperti contoh pada tahap eksplorasi !
- ${}_8\text{Be}$ dan ${}_{16}\text{S}$



- ${}_3\text{Li}$ dan ${}_{17}\text{Cl}$



- $^{20}_{20}\text{Ca}$ dan $^{35}_{35}\text{Br}$

