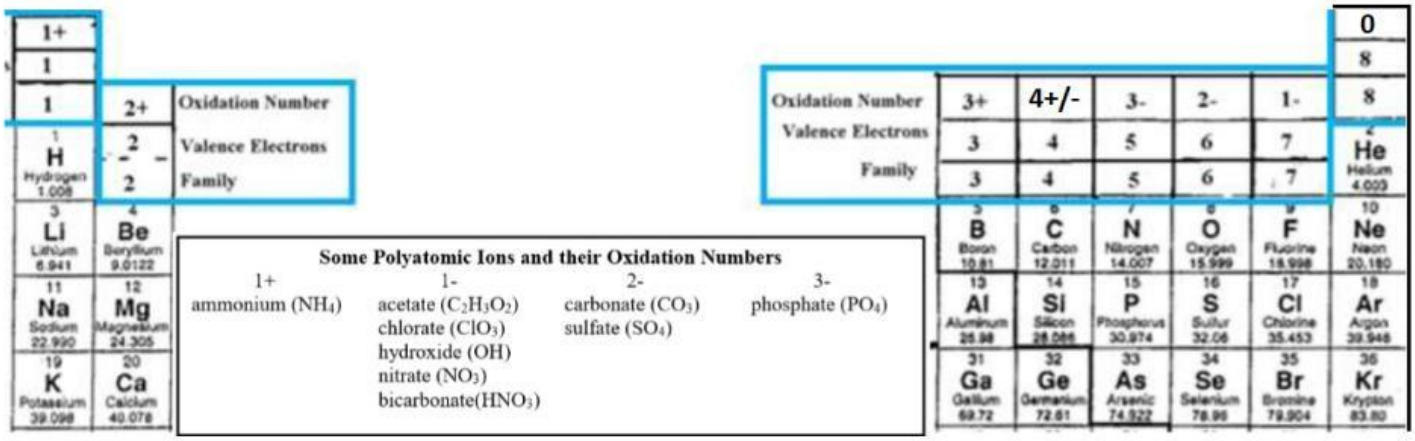


8. The oxidation number of an ion is the number of electrons an atom will lose to become a **negative positive** ion or gain to become a **negative positive** ion and have a full outer energy level of electrons.

Combine the elements or polyatomic compounds and write the formula and the name

	Symbols & Oxidation #	Formula	Name of Compound
9. Sodium and phosphate			
10. Magnesium and oxygen			
11. Potassium and phosphorus			
12. Ammonium and hydroxide			
13. Sodium and chlorine			
14. Beryllium and bromine			

LIVEWORKSHEETS