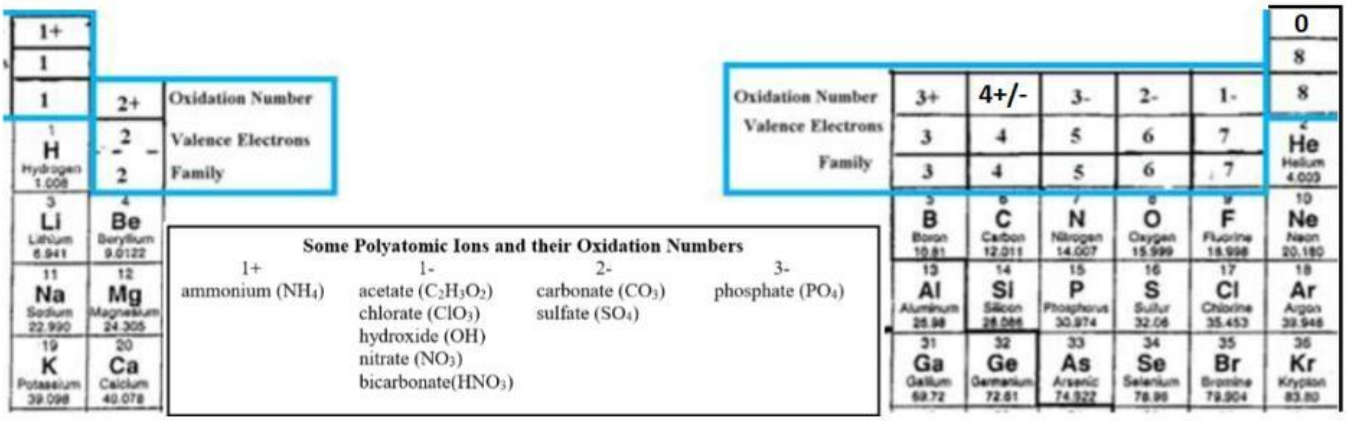


8. The oxidation number of an ion is the number of electrons an atom will lose to become a **negative positive** ion or gain to become a **negative positive** ion and have a full outer energy level of electrons.

Combine the elements or polyatomic compounds and write the formula and the name

	Symbols & Oxidation #	Formula	Name of Compound
9.Sodium and bicarbonate			
10. Lithium and nitrogen			
11. Beryllium and hydroxide			
12. Magnesium and chlorine			
13. Calcium and nitrate			
14. Potassium and acetate			
15. Sodium and phosphate			
16. Beryllium and bromine			
17. Aluminum and sulfate			
18. Sodium and carbonate			

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