MINI-TEST UNIT 2.

MATERIALS: PROPERTIES AND CHANGES

Question 1. Alex planed a test to compare boiling points of four substances: pure water, vinegar, cooking oil and ethanol

- a. Circle the correct answer
- i) In this test, the boiling points of different substances is

control variable dependent variable independent variable

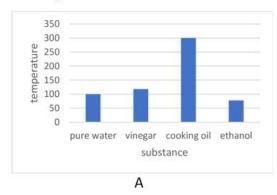
[1]

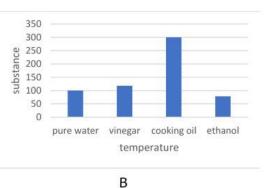
ii) Choose two materials and equipment will Alex need

thermometer cylinder force meter rod

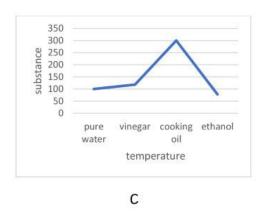
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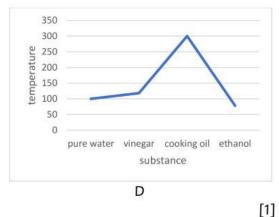
iii) This is his result: Pure water boils at 100 °C, vinegar boils at 118 °C and cooking oil boils at 300 °C, ethanol boils at 78°C. Choose the correct bar graph showing this result











iv) If Alex boils put these four substances with 90°C, which substances can boil?

Pure water ethanol cooking oil vinegar [1] Explain your choice[1] b. Explain why different substance boil at different temperatures.[1]

Question 2. Tick the characteristics of boiling but not evaporation

- a. Happen at any temperature
- b. Happen only at boiling point
- c. Has bubbles
- d. Occurs throughout the liquid
- e. Occurs only on the surface of the liquid
- f. Particles escape the surface of the liquid at all directions
- g. Particles escape the surface of the liquid at one direction upward

[4]





Question 3. a. Explain why the pot is made from metal [1] b. Explain why the handles of pot is made from plastic [1] c. Suggest one other material for making the handles. [1] Question 4. Kim Khanh set up a circuit, but the wire is broken. She touches the free ends of the wire to a rod. a. Suggest suitable material using for making the rod [1] b. Explain your choice in part a.

Question 5. Tick ALL irreversible changes

- a. melting chocolate in a hot pan
- b. frying an egg
- c. dissolving sugar in water
- d. burning matchsticks on a fire
- e. rusting on a pan

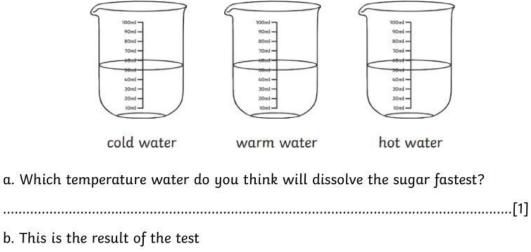
[3]

Nguyen Thi Nga - Checkpoint, IGCSE Science teacher

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Question 6. Ms Nga put 3 spoonfuls of sugar in a jar of cold/warm/hot water and time how long it takes to dissolve. Carry out 3 tests, one for each temperature and time how long it takes for the sugar to dissolve.

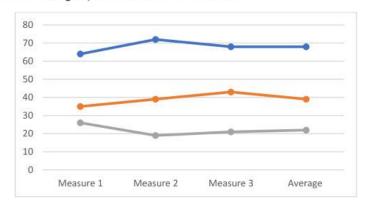


| Time (second) | Cold water | Warm water | Hot water |
|---------------|------------|------------|-----------|
| Measure 1 | 64 | 35 | 26 |
| Measure 2 | 72 | 39 | 19 |
| Measure 3 | 68 | 43 | 21 |
| Average | 68 | 39 | 22 |

| b. In which cond | lition, sugar dissolved | l fastest? | |
|------------------|-------------------------|---------------------------------|--------|
| Cold water | warm water | hot water | |
| | | | [1] |
| c. In which cond | lition, sugar dissolved | slowest? | |
| Cold water | warm water | hot water | |
| | | | [1] |
| d. Why Ms Nga i | measured time in thro | ee times and calculated the ave | erage? |
| | | | [1] |



e. Ms Nga drew a line graph the show the result



State TWO incorrect points in this graph.

| The first incorrect is | [1] |
|--|-------|
| The second incorrect is | [1] |
| Question 7. When we mix vinegar and baking soda, we can see bubbles of dioxide. | arbon |
| a. Is this a chemical change or physical change? Explain your answer | |
| | |

| This is achange | [1] |
|--------------------------------------|-----|
| Because | |
| b. Name the reactants in this change | |
| c. Name the product in this change | [2] |
| c. Nume the product in this change | [1] |
| d. State the evidence of this change | |

