



English – Class work



Look and Match. (Elige la opción correcta)

The number of protons in the nucleus of the atom.	CARBON 6 C 12.01	Usually from a Greek or Latin word for the element or a substance containing the element.
The average mass of the atoms in an element.		Short-hand abbreviation for the element name.

Parts of an ATOM. (Arrastra los nombres a la definición correcta.)

Electron

Proton

Neutron

- _____ Positively charged particle in the nucleus of the atom.
The number of protons in an atom's nucleus is the atomic number.
- _____ Negatively charged particle surrounding the nucleus of the atom.
The number of electrons surrounding the nucleus of an atom is equal to the number of protons in the atom's nucleus.
- _____ Particle in the nucleus that has about the same mass as a proton but has no charge. For the atoms of the first 20 elements, the number of neutrons is either equal to or slightly greater than the number of protons.