|   | Date:   |
|---|---|
| se the periodic table   | to help you answer these questions.                 |
| How many differen   | nt elements are present in the following compounds: |
| a KCI   | b NaBr  |
| c Cal <sub>2</sub>  | d CuCl <sub>2</sub>                                 |
| e Al <sub>2</sub> O <sub>3</sub>  | f NH <sub>4</sub> OH                                |
| g NaNO <sub>3</sub>   | h NH <sub>4</sub> NO <sub>3</sub>                   |
| i CuCO <sub>3</sub>   | j Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>   |
| Name each eleme   | ent present in the following ionic compounds:       |
| a NaCl  | and present in the following following following.   |
|   |   |
| b CaCO <sub>3</sub>   |   |
| b CaCO <sub>3</sub>   |   |
| g Fe <sub>2</sub> O <sub>3</sub>  |   |
| A Fe <sub>2</sub> O <sub>3</sub>  |   |
| A Fe <sub>2</sub> O <sub>3</sub>  |   |
| £ Fe <sub>2</sub> O <sub>3</sub><br>£ NH <sub>4</sub> Br<br>£ Mg(OH) <sub>2</sub><br>£ KI |   |

- 3 Name each element present in the following ionic compounds:
  - a NaF
  - h RbBr
  - & BaSO.
  - d V<sub>2</sub>S<sub>3</sub>
  - & WC
  - f Na<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>
  - g BF<sub>a</sub>
  - h KMnO<sub>4</sub>