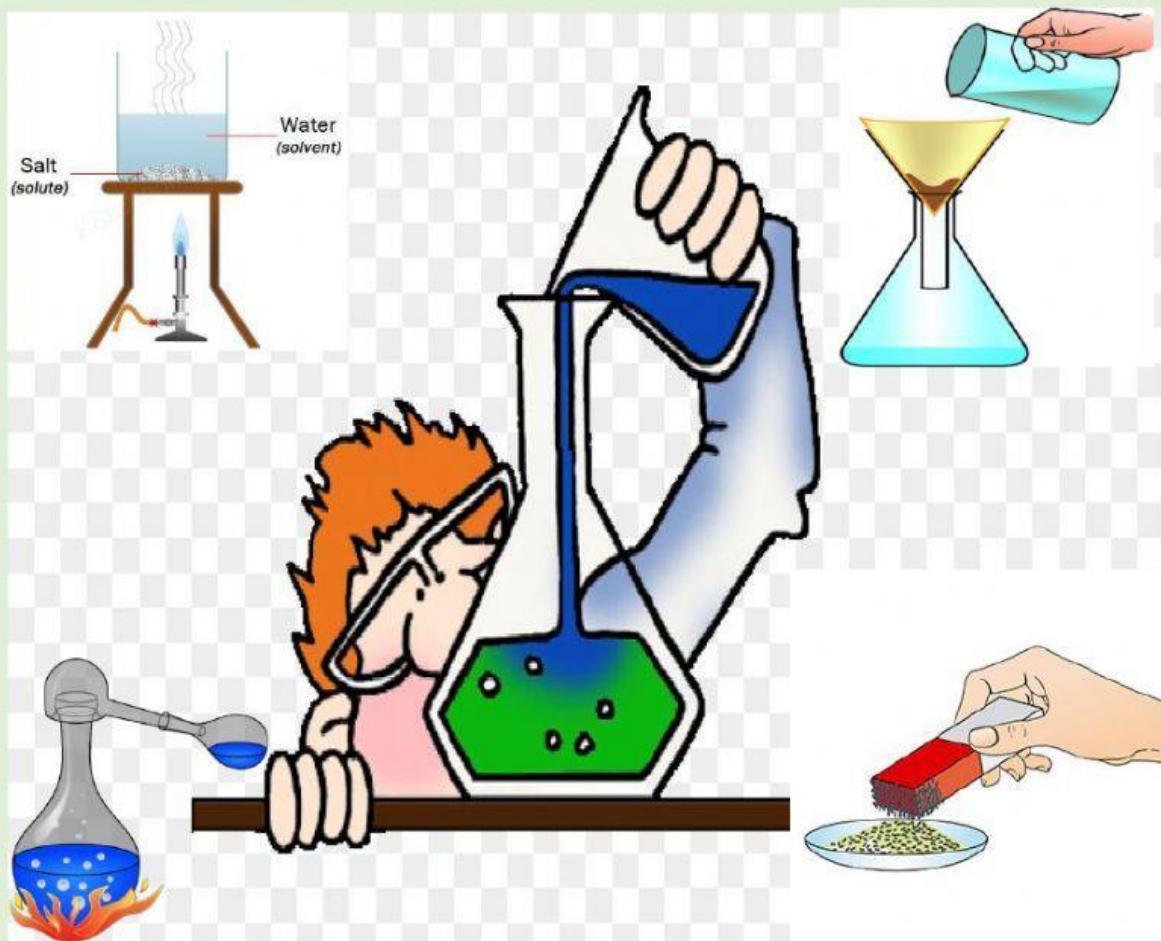




## SCIENCE

### CHAPTER 6 -MATTER AND ITS CHANGES LESSON 4- MIXTURES



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## MIXTURES

- ❖ A mixture is a **physical combination** of two or more kinds of matter.
- ❖ It is a **physical combination**.
- ❖ Mixtures can be separated by **physical means**.
- ❖ **Everyday Mixtures**

A salad is usually a mixture of lettuce, tomatoes, and other foods. The foods that go into a salad might be chopped up. However, they are still the same kinds of food.

You probably see mixtures every day. Some breakfast cereals are mixtures of solids. If you add milk, you get a mixture of solids and a liquid. Many products, such as food, drinks, and clothing, are made from mixing different kinds of matter.

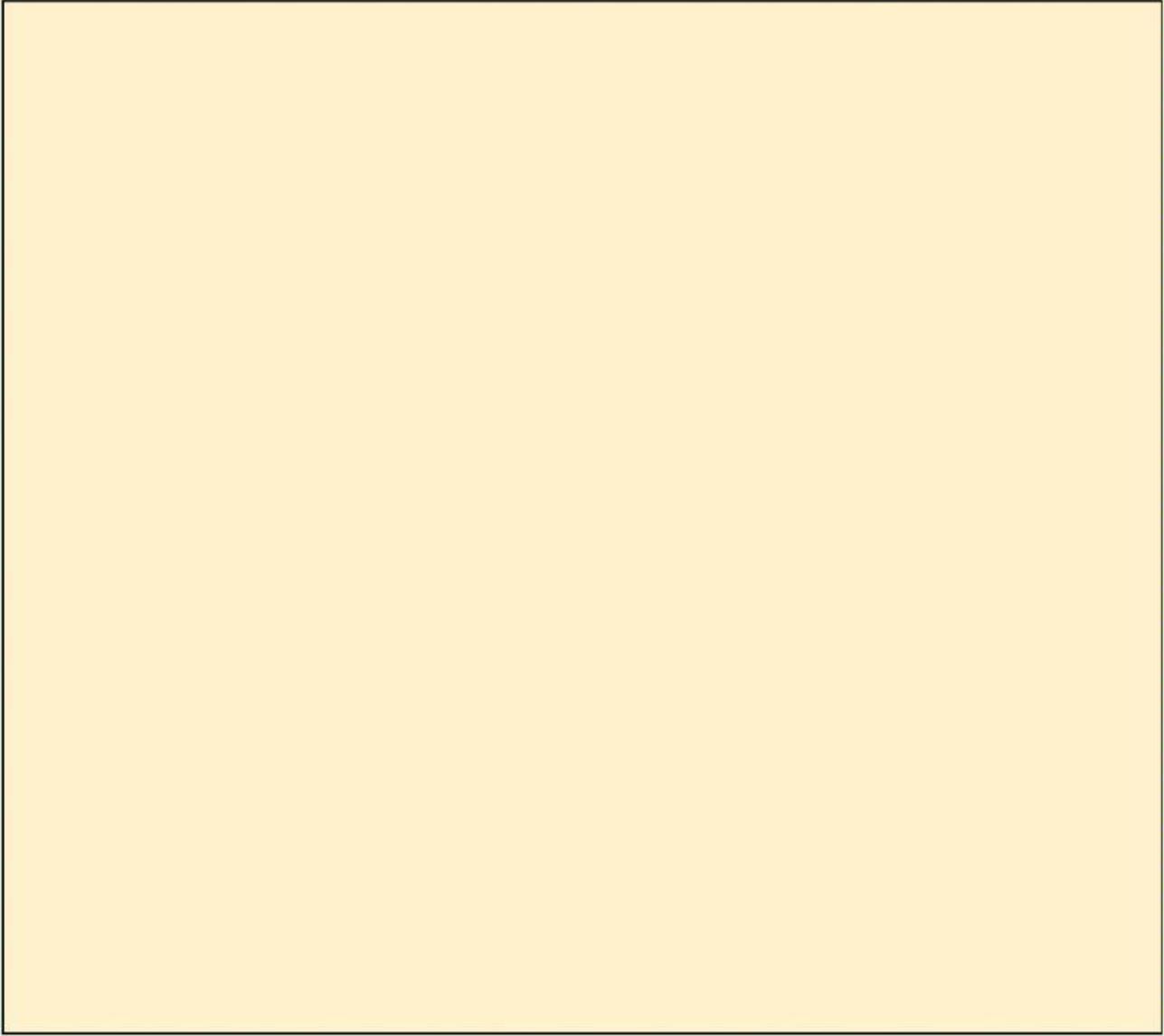
### **Examples of Mixtures**



## SOLUTIONS ARE SPECIAL TYPE OF MIXTURES

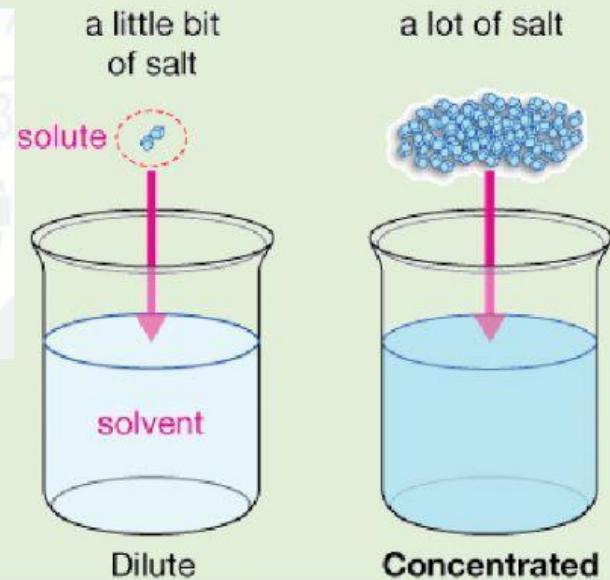
Some solids mix easily with liquids. If you mix salt into water, the salt will break up. Salt water is a solution. A **solution** is a mixture in which two or more substances are blended completely.

**WATCH VIDEO ABOUT MIXTURES AND SOLUTIONS**



## SOLUTION LIMITS

When there is just a little sugar in water, it is called a dilute sugar solution. This water is not very sweet. Adding more sugar makes the solution more concentrated and also sweeter.

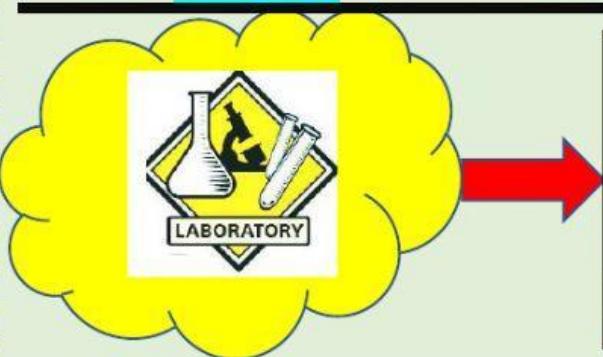


## SATURATED SOLUTION

❖ When no more solute (salt/ sugar) can be added to the solution at certain temperature, it is called saturated solution.

Eventually, a solvent can't hold anymore!

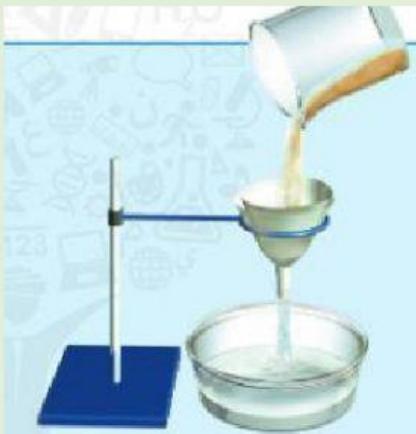
A saturated solution **CANNOT** hold any more solute at a certain temperature.



- **Click** here to do lab about solution.
- **Click** here to do lab about saturated solution

## HOW TO SEPARATE MIXTURES

### FILTRATION



Sand particles cannot pass through the pores in the filter, but water particles can easily.

### EVAPORATION



Dissolving sugar in water separates it from sand. Filtering and evaporating the water recovers the sugar.

## **MAGNETISM**



Iron is attracted to a magnet, but sand is nonmagnetic.

## **BUOYANCY**



In water, the low-density sawdust floats while the high-density sand sinks.



## IMPORTANT!

### IMPORTANT VIRTUAL LAB FOR PRACTICE

- **Click** here to learn how to separate mixtures using virtual lab
- **Click** here to do practice separation of mixtures using game.

### IMPORTANT USE OF MIXTURES

#### **ALLOY**

- **Melted metals are mixed to make alloy.**

Sometimes when mixed into alloys, the properties of metals seem to blend together. For example, pure copper is soft and flexible and pure zinc is hard and brittle. Brass, an alloy of copper and zinc, is hard but still flexible. It is used to make musical instruments.

## **DISTILLATION**

- It is a process of separating liquids by evaporation and condensation.
- If two liquids have different boiling points, we can use this method.

## QUESTIONS FROM BOOK