Name: _____

Balancing Chemical Equations Extra Credit (10pts)

You have to show your work to receive full credit for each equation.

1. _____ H_2 + _____ N_2 → _____ NH_3 (2pts)

Reactants	Products

2. _____ Al_2O_3 → _____Al + _____ O_2 (2pts)

Reactants	Products

3. _____ C_2H_6 + _____O_2 → _____CO_2 + _____H_2O

Reactants	Products

4. _____H_2O_2 → _____O_2 + _____H_2O

Reactants	Products

5. _____Fe + _____O_2 → _____Fe_2O_3

Reactants	Products