

- \_\_\_\_ 12. When a solute is added to a solvent, (ex. salt in water) the freezing point of the solution is
- higher than the freezing point of the pure solvent.
  - lower than the freezing point of the pure solvent.
  - the same as the freezing point of the pure solute.
  - the same as the freezing point of the pure solvent.
- \_\_\_\_ 13. Which is a characteristic property of acids?
- Acids turn blue litmus paper red.
  - Acids turn red litmus paper blue.
  - Acids taste bitter.
  - Acids do not react with metals.
- \_\_\_\_ 14. Strong acids can produce designs on metal printing plates because they
- turn red litmus paper blue.
  - conduct electricity.
  - are corrosive.
  - react with limestone.
- \_\_\_\_ 15. A substance that tastes bitter, feels slippery, and turns red litmus paper blue is a(n)
- acid.
  - base.
  - indicator.
  - salt.
- \_\_\_\_ 16. You are most likely to find a base in
- brick and metal cleaners.
  - a car battery.
  - fruit juice.
  - household cleaners.
- \_\_\_\_ 17. Any substance that forms hydrogen ions ( $H^+$ ) in water is a(n)
- acid.
  - base.
  - indicator.
  - salt.
- \_\_\_\_ 18. In water, bases form
- hydroxide ions.
  - hydrogen ions.
  - hydrogen gas.
  - oxide ions.
- \_\_\_\_ 19. Acids naturally present in food are safe to eat because they usually are
- concentrated.
  - dilute.
  - strong.
  - weak.
- \_\_\_\_ 20. The pH scale measures the
- strength of an acid.
  - solubility of an acid.
  - concentration of hydrogen ions.
  - concentration of an acid.
- \_\_\_\_ 21. You can use indicators to
- find the concentration of a solution.
  - test for conductivity.
  - find the pH of a solution.
  - find out if a solution is saturated.
- \_\_\_\_ 22. If a solution has a pH of 9, the solution is
- acidic.
  - basic.
  - neutral.
  - saturated.
- \_\_\_\_ 23. Normal rainfall is slightly acidic, which means its pH must be
- less than 2.
  - between 5 and 7.
  - between 2 and 4.
  - between 7 and 9.
- \_\_\_\_ 24. Neutralization is a reaction between
- an acid and a base.
  - an acid and a metal.
  - a base and a salt.
  - salt and water.
- \_\_\_\_ 25. A neutralization reaction produces
- a neutral solution.
  - an acid and a base.
  - water and a salt.
  - table salt.