تنبأ باشار الانتربي للنظام في التغيرات الفيزيائية و الكيميائة التالية

1.	$2KClO_3(s) \rightarrow 2KCl(s) + 3O_2(g)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
2.	$H_2O(l) \rightarrow H_2O(s)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
3.	$N_2(g) + 3H_2(g) \rightarrow 2NH_3(g)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
4.	$NaCl(s) \rightarrow Na^+(aq) + Cl^-(aq)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
5.	$KCl(s) \rightarrow KCl(l)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
6.	$CO_3(s) \rightarrow CO_2(g)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
7.	$H^+(aq) + C_2H_3O_2^-(aq) \rightarrow HC_2H_3O_3(l)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
8.	$C(s) + O_2(g) \rightarrow CO_2(g)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
9.	$H_2(g) + Cl_2(g) \rightarrow 2HCl(g)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
10.	$Ag^+(aq) + Cl^-(aq) \rightarrow AgCl(s)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
11.	$2N_2O_5(g) \rightarrow 4NO_2(g) + O_2(g)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
12.	$2\,Al(s) + 3I_2 \rightarrow 2\,AlI_3$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
13.	$H^+(aq) + OH^-(aq) \rightarrow H_2O(l)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
14.	$2NO(g) \rightarrow N_2(g) + O_2(g)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$
15.	$H_2O(g) \rightarrow H_2O(l)$	$\Delta S > 0$	$\Delta S = 0$	$\Delta S < 0$