Name: _____ Homeroom Teacher:

Ivanic.	MATTER UNIT ASSESSMENT REVIEW
1. List at least 10	1.
physical	2.
properties of	3.
matter.	4.
macter.	5.
	6.
	7.
	8.
	9.
	10.
2. Give five examples	1.
of how matter can	2.
change physically.	3.
anange pinyereamy.	4.
	5.
3. Give five examples	1.
of how matter can	2.
change chemically .	3.
	4.
	5.
4. Define what a	Definition:
mixture is and give	
3+ examples.	Ex:
	1.
	2.
	3.
5. Describe ways to	Wood & rice:
separate these	Wood & Hoo!
mixtures:	
illixtures.	Water & sand:
	water & sariu.
	Water & salt:
	Iron fillings & salt:
	P0
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Name:	Homeroon	ı Teacher:	
Describe what a solution is while including examples.			
7. Why is a solution also considered a special kind of			
mixture?			
8. Explain solubility.			
 Do all properties change when one substance dissolves into another? Explain. 			
10. Give 2+ examples of substances that would dissolve in water & how you would separate them.			
11. When you are dissolving water from sugar or salt, why doesn't the mass of the substances change?			

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12. How is the mass and volume of an object related to whether it will sink or float in water?				
Provide an example utilizing <i>numbers</i> .				
13. List the tools we used during the density blocks experiment and how they were used.				
14. Describe a solid & draw an example of what their particles look like.				
15. Describe a liquid & draw an example of what their particles look like.				
16. Describe a gas & draw an example of what their particles look like.				
17. Define volume & give an example.18. List the tools we use for volume.				

Name:	Homeroom Teacher:			
19. Define buoyancy& give an example				
20. Define mass & give an example				
21. List the tools we use for mass.				
22. Define weight & give an example				
23. List a tool we use for weight.				
24. Define density & give an example				
 Explain the process of evaporation. 				
26. Draw the solid, liquid, gas cycle. Be sure to include where heat increases or decreases in your cycle and the temperatures of freezing, melting, and boiling points.				