

Mystery Elements in a Mock Periodic Table

Procedure: Using your knowledge of valence electrons and the periodic table, fill in the mystery elements in their proper spaces.

- Element Ph is a Noble gas with its 4th energy level filled with 8 electrons. Family _____ Period _____ Ph
- Element Bc has its 2nd energy level filled with 8 valence electrons. Family _____ Period _____ Bc
- Element X has only 1 valence electron and is in the same period as element Ph. Family _____ Period _____ X
- Element Tq has 2 valence electrons and two energy levels. Family _____ Period _____ Tq
- Element Ra is in the same period as element Bc and has 1 valence electron. Family _____ Period _____ Ra
- Element Qr is a Nobel gas with the smallest atomic number. Family _____ Period _____ Qr
- Element St is also a Nobel gas. Family _____ St
- Element Rn has 1 valence electron and three energy levels. Family _____ Period _____ Rn
- Element Az has 1 valence electron and the highest atomic number of that family. Family _____ Az
- Element Z has 2 valence electrons and the highest atomic number in the family. Family _____ Z
- Element B has 7 valence electrons and only 2 energy levels. Family _____ Period _____ B
- Element O has 6 electrons in its outer level and 3 energy levels. Family _____ Period _____ O
- Element Sp has 7 valence electrons and in the same period as Rn. Family _____ Period _____ Sp
- Element Mk is in the same family as element B. Family _____ Period _____ Mk
- Element H is a metal with 2 valence electrons and 3 energy levels. Family _____ Period _____ H
- Element Hp has the lowest atomic number of all the elements. This mean it's Atomic number is _____ Hp
- Element Pt is a metalloid in the same period as X, it has 5 valence electrons. Family _____ Period _____ Pt
- Element J is a metalloid with the lowest atomic number of all metalloids. It's atomic number is _____ J
- Element I and R are both in the same period as element Z, but R has a higher atomic number than I. I and R are on Period _____ R goes farther to the _____ I R
- Element Tk has two valence electrons. Family _____ Period _____ Tk

Drag and drop the element symbol to the correct location

ENERGY LEVELS ARE THE PERIODS and go ACROSS

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|---|--|--|--|--|--|--|--|
| 1 | | | | | | | |
| 2 | | | | | | | |
| 3 | | | | | | | |
| 4 | | | | | | | |
| 5 | | | | | | | |

Families are valence electrons and go in columns

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|--|---|---|---|---|---|--|---|
| | 3 | 4 | 5 | 6 | 7 | | 8 |
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