

AP Biology Topic 1.3 – Introduction to Biological Molecules

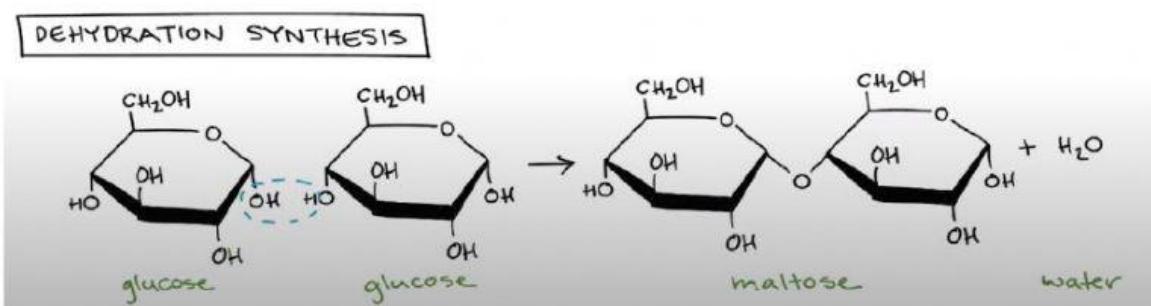
- Macromolecules are **constructed by the bonding of smaller molecules**

Structure of Monomers and Polymers – Use the letter 'O' to make the structures below

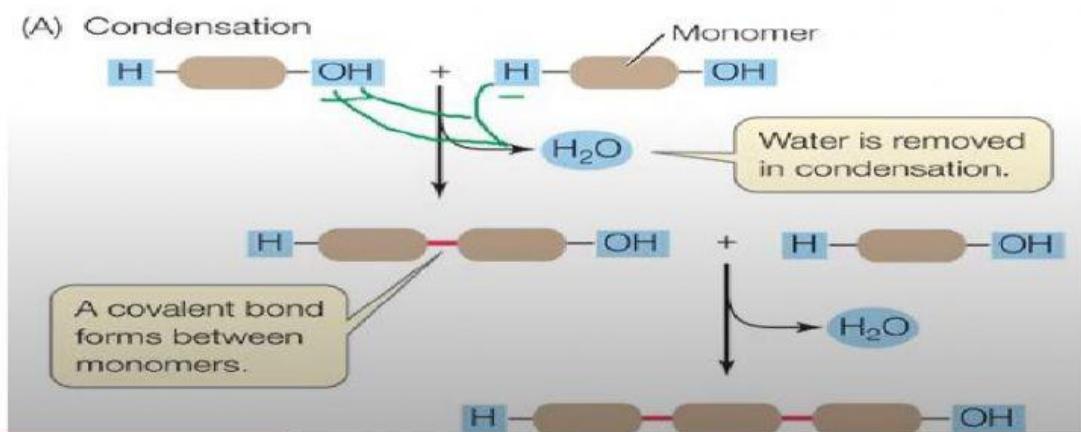
Monomer draw picture in space below: a **monomer** is a small molecule

Polymer draw picture in space below: a pattern of

- Polymers are formed through reactions.
- Dehydration synthesis means “ ”
- It is also considered to be a since two molecules are condensed into one larger molecule with the loss of a smaller molecule ()

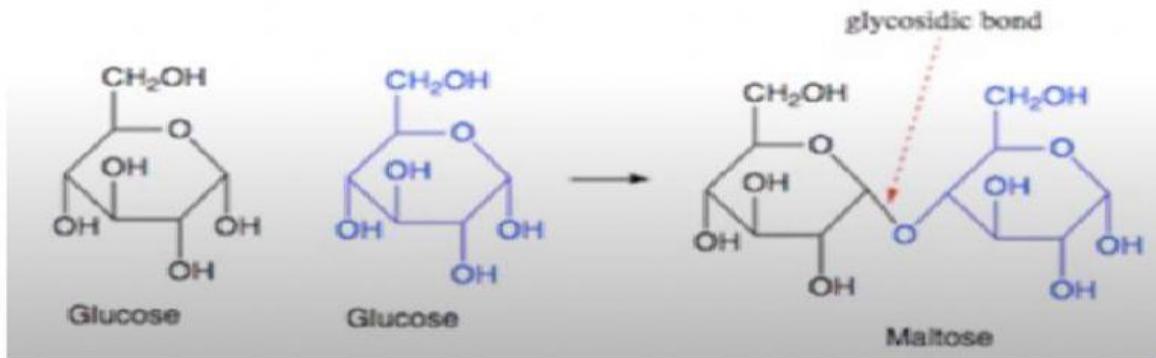


- In a dehydration synthesis reaction, the **carboxyl group** of one monomer combines with the **hydroxyl group** of another monomer, releasing a molecule of **water**.
- The monomers share **carboxyl groups** and form **polymers**.
- As additional monomers join, this chain of repeating

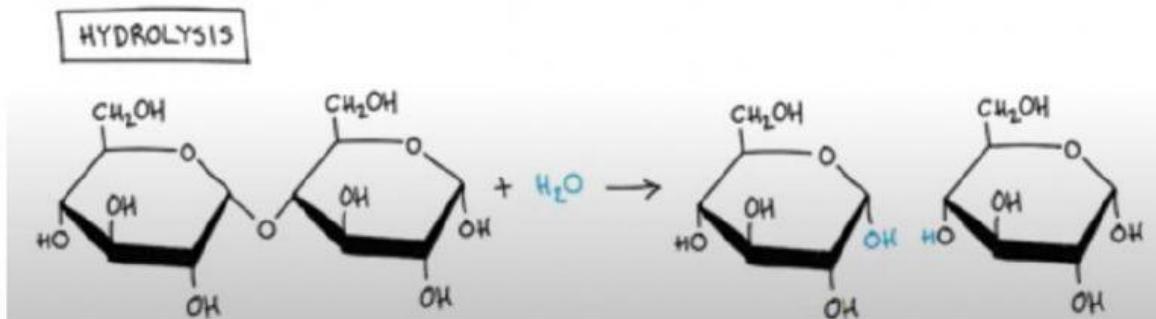


- Different types of are formed during synthesis in different macromolecules:

Carbohydrate	Lipid	Protein	Nucleic Acid



- Polymers are broken down through reactions.
- Hydrolysis means “ ”
- During these reactions, the is broken into two components: one part gains a and the other gains a from a water molecule.



- What is the first thing that happens during hydrolysis?
- What does this do to a covalent bond?
- Identify and explain the process we use to build monomers.