I. Use the Periodic table to write the symbol of each "pure" elements

Example: Calcium	the symbol is Ca
1. Silicon	2. Barium
3. Carbon	4. Titanium
5. Copper	6. Gold
7. Chlorine	8. Mercury
9. Potassium	10.Aluminum
11.Boron	12.Oxygen

II. Read

Elements

All elements are located on the Periodic table—
They are made of one kind of atom
(Example: Nitrogen (N), is 7th element in the Periodic Table)

Compounds

Compounds are made of two or more elements combined chemically (example: H₂O is made of Hydrogen and Oxygen, combined chemically)

Write element or compound

13.	Au		
14.	CaH2 Calcium hydride		
15.	Ga	<u> </u>	
16.	Cu	-	
17.	H₂O		

18.	Rf
19.	H2SO4 (sulfuric acid)
20.	HCl (Hydrogen chloride)
21.	O2
22.	co
23.	CO ₂
24.	How about Sea Water? —it is a mixture of many compounds

III. Final question

A compound H₂SO₄ (sulfuric acid) is made of 3 different elements: H (hydrogen), S (Sulfur), and O (Oxygen)

25. How many different elements Na H C O₃ (baking soda) is made of?

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