












CHEMISTRY

DALTON'S ATOMIC THEORY

1. All matter is made up of small particles called 
2. Atoms of the same element are always 

3. Atoms of different elements combine in simple whole number ratios to form 
4. The word 'atom' is derived from a Greek word 'a-tomio' which means 
5.  are neither created nor destroyed in a chemical reaction.
6. When carbon and oxygen combine in 1:1 ratio by volume, then they form 
7. The concept of 'anu' and 'parmanu' was proposed by 
8. According to Kanad, matter is composed of very small particles known as anu. Here 'anu' represents 
9. The number of atoms present in one molecule of ozone is 
10. The number of hydrogen atoms combined with one atom of oxygen in the formation of water is 



AIMS - INDIA

THE BEST IIT-JEE/NEET & OLYMPIAD COURSE PROVIDER IN INDIA

