

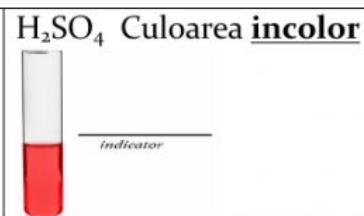
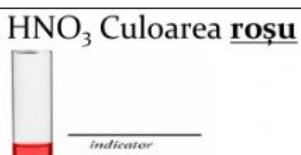
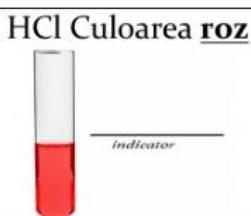
Tema: Acizii

Varianta I

Timp: 10 min

1. Completează și egalează ecuațiile reacțiilor de mai jos:

- a) $\text{Al} + \text{HCl} \rightarrow \dots + \dots$
- b) $\text{MgO} + \dots \rightarrow \text{Mg}(\text{NO}_3)_2 + \dots$
- c) $\text{CuO} + \dots \rightarrow \text{CuSO}_4 + \dots$
- d) $\text{Ba}(\text{NO}_3)_2 + \dots \rightarrow \text{BaSO}_4 \downarrow + \dots$
- e) $\text{Fe(OH)}_3 + \text{HCl} \rightarrow \dots + \dots$
- f) $\text{HCl} + \dots \rightarrow \text{AgCl} \downarrow + \dots$



Denumirea formulei chimice:

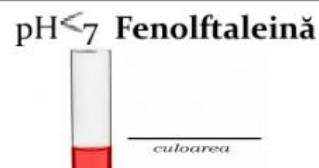
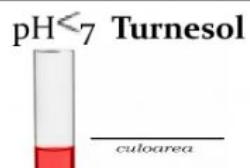
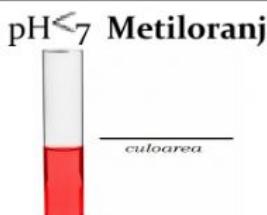
HCl _____; HClO₃ _____; H₂SO₄ _____.**Tema: Acizii**

Varianta II

Timp: 10 min

1) Completează și egalează ecuațiile reacțiilor de mai jos:

- a) $\text{Pb}(\text{NO}_3)_2 + \dots \rightarrow \text{PbI}_2 \downarrow + \dots$
- b) $\text{CaO} + \text{HNO}_3 \rightarrow \dots + \dots$
- c) $\text{Na}_2\text{O} + \dots \rightarrow \text{Na}_2\text{SO}_4 + \dots$
- d) $\text{BaCl}_2 + \text{H}_2\text{SO}_3 \rightarrow \dots + \dots$
- e) $\text{Fe(OH)}_3 + \text{H}_2\text{SO}_4 \rightarrow \dots + \dots$
- f) $\text{HNO}_3 + \text{Mg(OH)}_2 \rightarrow \dots + \dots$



Denumirea formulei chimice:

HBr_____; HClO₄_____; HNO₃_____.