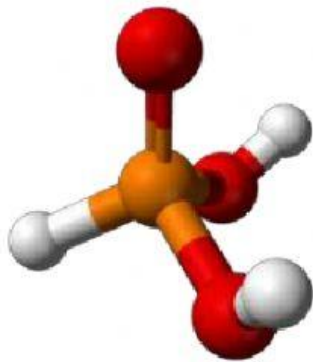
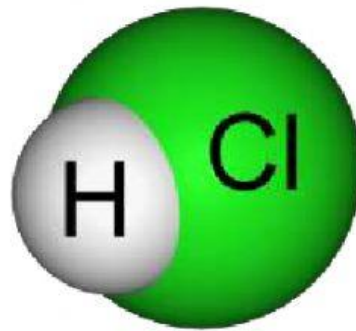


## Nomenclatura tradicional hidrácidos y oxácidos

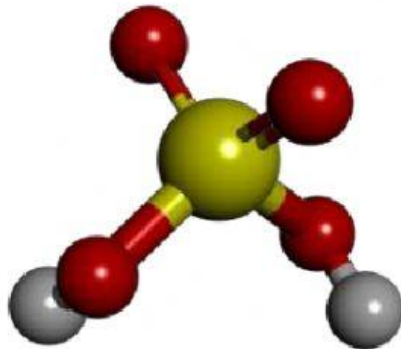
1.  $\text{H}_3\text{PO}_3$



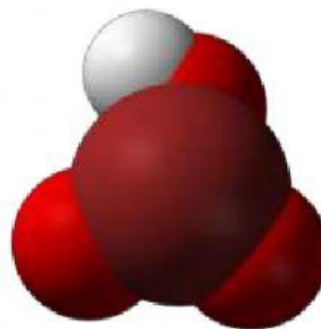
2.  $\text{HCl}$



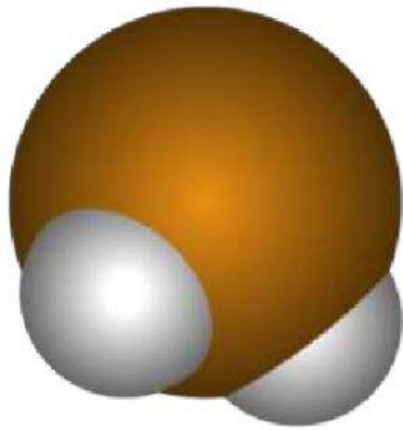
3.  $\text{H}_2\text{SO}_4$



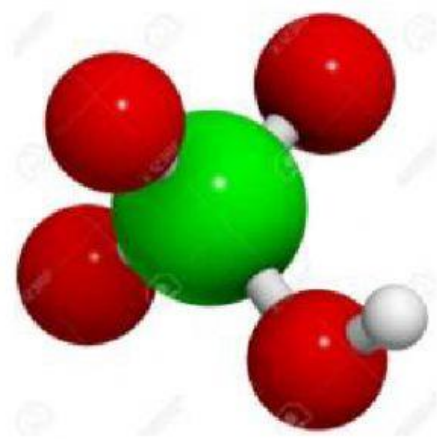
4.  $\text{HBrO}_3$



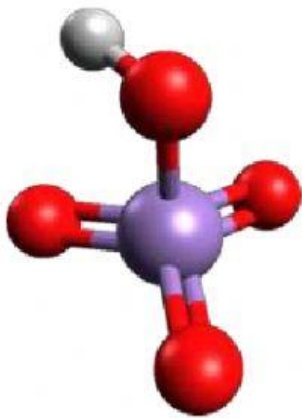
5. H<sub>2</sub>Te



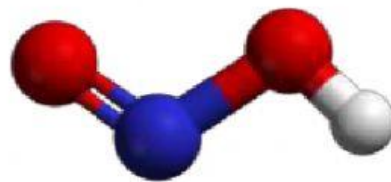
6. HClO<sub>4</sub>



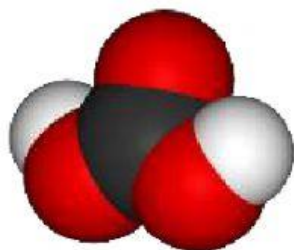
7. HMnO<sub>4</sub>



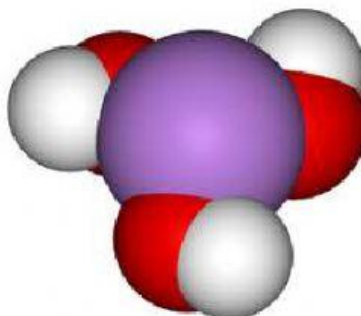
8. H<sub>2</sub>NO<sub>2</sub>



11.  $\text{H}_2\text{CO}_3$



12.  $\text{H}_3\text{AsO}_3$



13.  $\text{H}_4\text{P}_2\text{O}_7$

