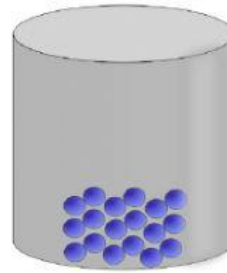
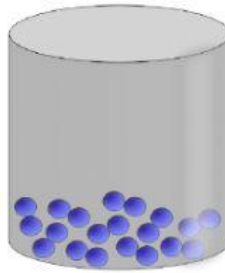
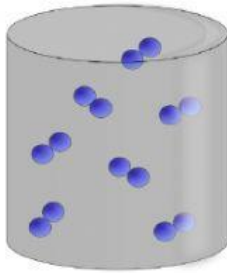


Name: \_\_\_\_\_ Date: \_\_\_\_\_

### States of Matter

Label the state of matter shown by each group of particles.



\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

Complete the sentences about the properties of each state of matter. Use the words from the brackets.

#### Solids

( compressed, same, volume, close, cutting )

The particles in solids are very \_\_\_\_\_ together.

The shape of a solid stays the \_\_\_\_\_.

You can change the shape of a solid by \_\_\_\_\_ it.

Solids have a fixed \_\_\_\_\_ so they cannot be \_\_\_\_\_.

## **Liquids**

**( container, poured, compressed, particles, shape )**

Liquids can be \_\_\_\_\_ because the \_\_\_\_\_ can move around each other.

Like solids, their volume is fixed and they cannot be \_\_\_\_\_.

Liquids take on the \_\_\_\_\_ of their \_\_\_\_\_.

## **Gases**

**( shape, invisible, expand, odour, quickly )**

The particles in a gas move very \_\_\_\_\_.

Some gases are \_\_\_\_\_ and have no \_\_\_\_\_.

Just like liquids, gases do not have a \_\_\_\_\_ of their own.

With their particles are so far apart gases can \_\_\_\_\_ to use up all the available space.

Click and drag each item to the correct box.

ketchup

pumpkin

pillow

milkshake

bench

carbon dioxide

shampoo

oxygen

stream

smoke

flashlight

SOLIDS	LIQUIDS	GASES