

KONFIGURASI ELEKTRON

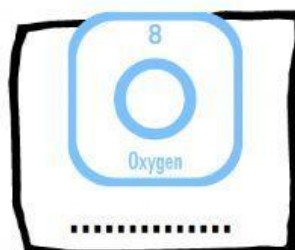
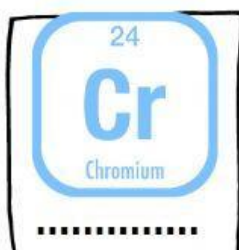
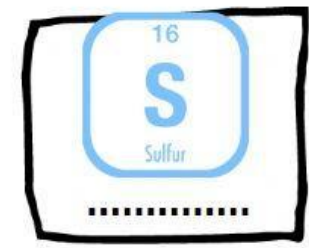
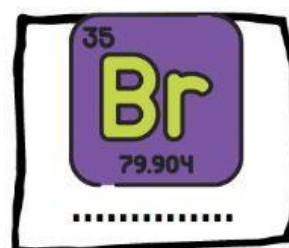
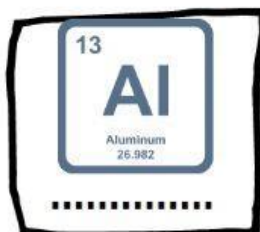
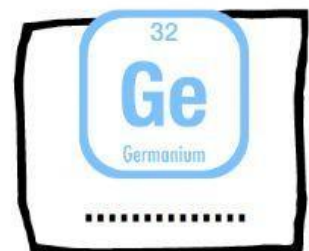
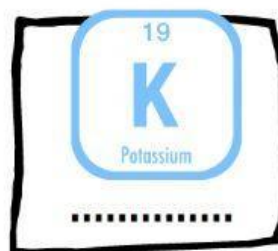
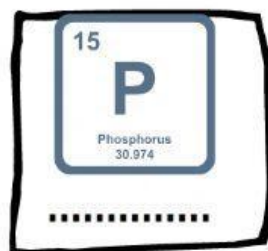
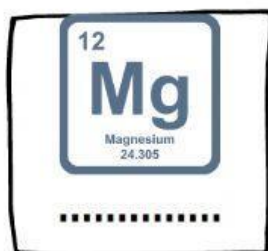
Konfigurasi elektron merupakan susunan elektron di dalam suatu atom. Konfigurasi elektron yang dipelajari saat ini berdasarkan kulit atom. Kulit atom terdiri dari:

- Kulit K (elektron maksimal 2)
- Kulit L (elektron maksimal 8)
- Kulit M ((elektron maksimal 18)
- Kulit N (elektron maksimal 32)

Contoh atom Kalsium dengan nomor atom 20, maka konfigurasi elektronnya 2, 8, 8, 2.

LATIHAN

Geser konfigurasi elektron di bawah ke gambar unsur yang sesuai!



2, 8, 8, 1
2, 8, 4

2, 8, 6
2, 8, 8, 6

2, 6
2, 8, 3

2, 8, 18, 7
2, 8, 5

2, 8, 18, 4
2, 8, 2