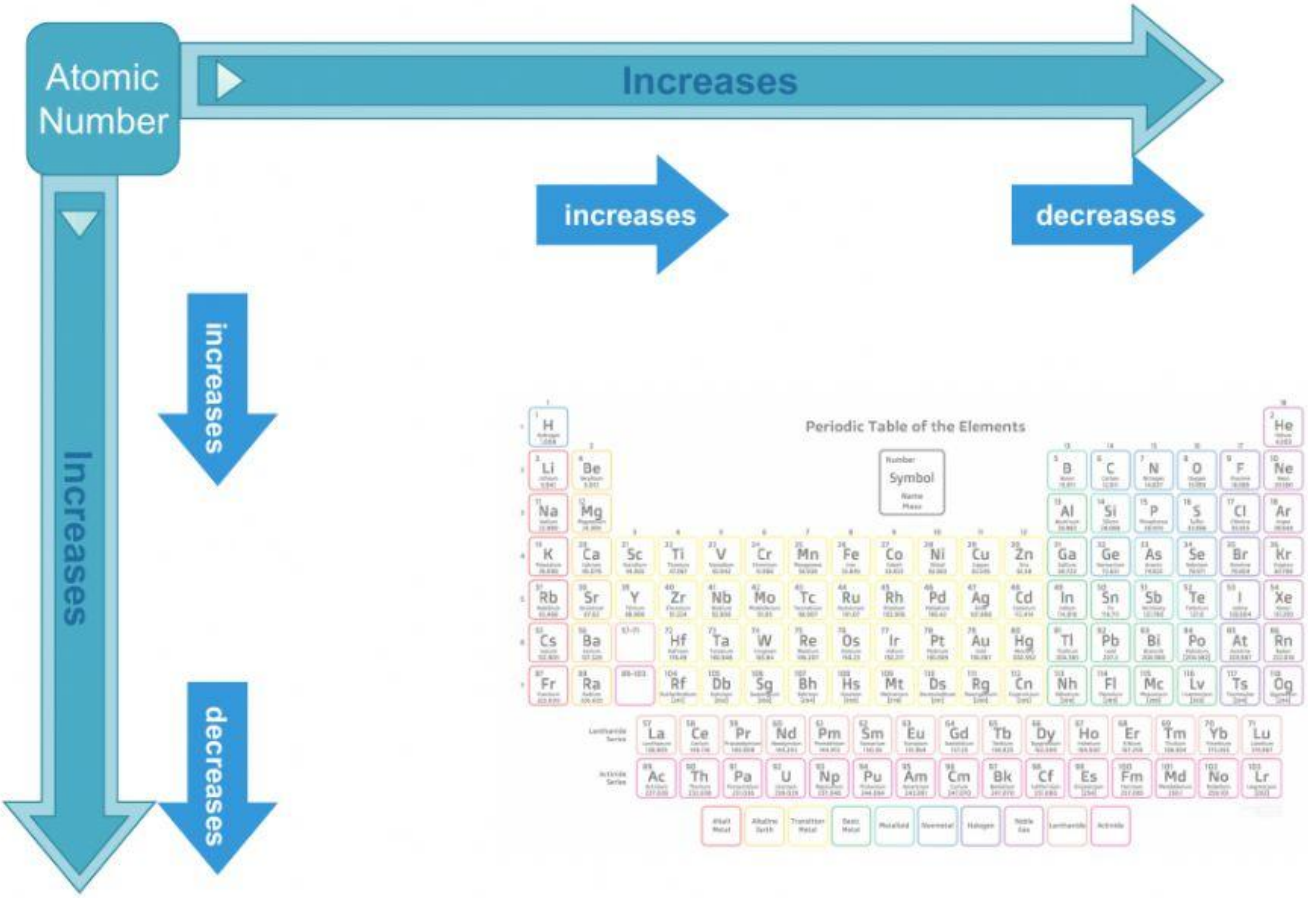
Graduation of elements properties in modern periodic table



- By increasing atomic number in group .....increase due to increasing number of energy levels.
- By increasing atomic number in period atomic size ..... due to increasing attraction force between nucleus and outer energy level's electrons.
- electronegativity increases by increasing atomic number in .....and decreases by increasing atomic number in .....
- strongest metals found at .....side of modern periodic table, while strongest non metals found at the .....side before inert gas
- the metalloid is found between .....element and .....element
- The measuring unit of atomic radius is .....
- metal oxide is .....oxide, while non-metal oxide is .....oxide

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