

1) Complete the sentence.

Mass

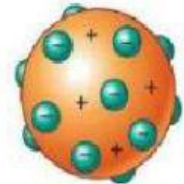
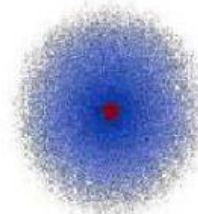
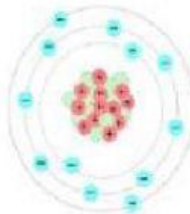
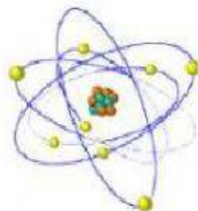
space

place

atoms

broken

- Matter: Is everything that surrounds you, has _____ and occupies a _____ in _____.
- Substances that are formed by the same _____, that cannot be _____ into anything simpler.



3) Line the names with their definition.

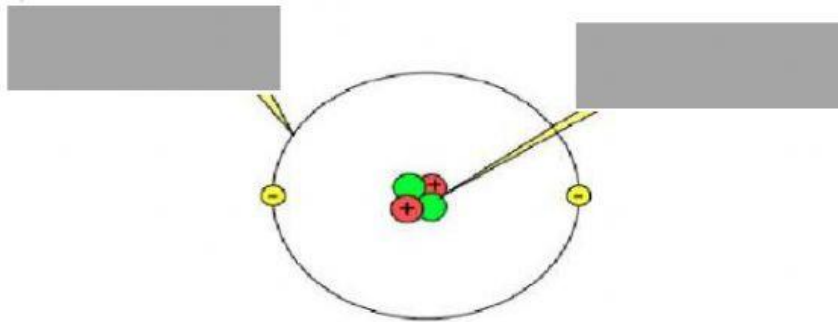
Enlace químico

Es la unión de dos o más átomos, debido a la fuerza de atracción o repulsión de los electrones.

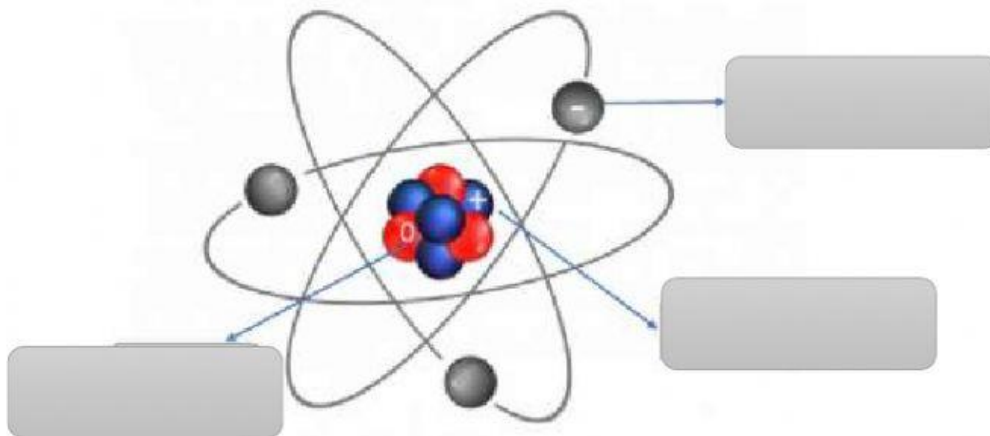
Fórmula química

Muestran los elementos que forman el compuesto y la proporción que existe entre ellos. Example: H₂O.

4) Fill in the blanks with the names of the parts of the atom.



5) Write the name of the subatomic particles where applicable.



6) Identify what each part of the image refers to.

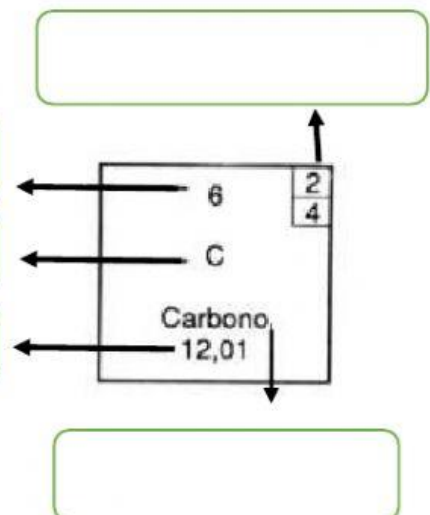


Diagram of the periodic table entry for Carbon (C). The entry is a box containing: '6' (atomic number), 'C' (symbol), 'Carbono' (name), and '12,01' (atomic mass). Arrows point from the empty boxes on the left to these four parts. An arrow points from the empty box on the right to the '2' and '4' in the top right corner of the box.

Distribución de
electrones en cada nivel.

Nombre

Número atómico

Símbolo

Masa atómica