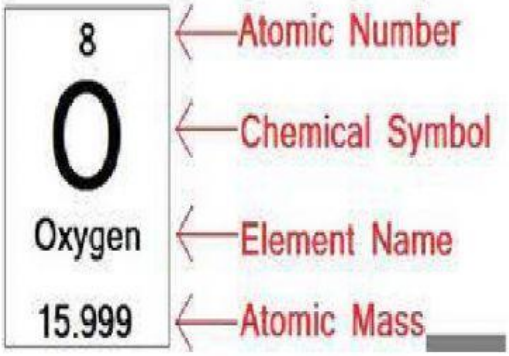
ATOMIC NUMBER AND ATOMIC MASS

REVISION

EXAMPLE:



Atomic Number = 8

Number of protons = 8

Number of electrons =8

Atomic mass = 16 (round up 15.999)

Number of neutrons= mass number – atomic number(protons)

= 16 – 8 = 8

Question 1

Fill in the blanks

Atomic Number =

Number of protons =

Number of electrons =

Atomic mass =

Number of neutrons=

LIVEWORKSHEETS

Question 2

Atomic Number =

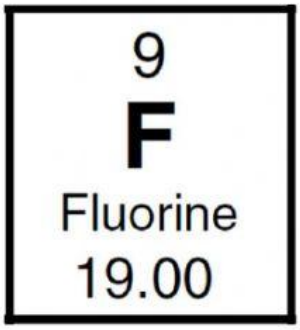
Number of protons =

Number of electrons =

Atomic mass =

Number of neutrons=

Question 3



Atomic Number =

Number of protons =

Number of electrons =

Atomic mass =

Number of neutrons=

LIVEWORKSHEETS