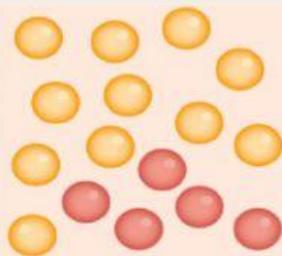
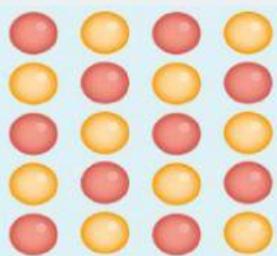


## Heterogeneous VS Homogeneous mixtures

Heterogeneous mixture



Homogeneous mixture



## SOLUTIONS

=Homogenous mixture

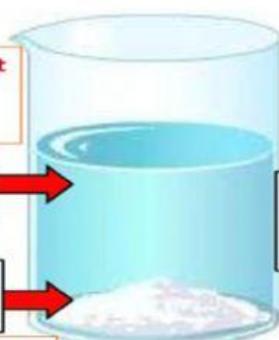
A solution is a mixture formed when one or more solutes dissolve in a solvent.

Substance present in the largest amount

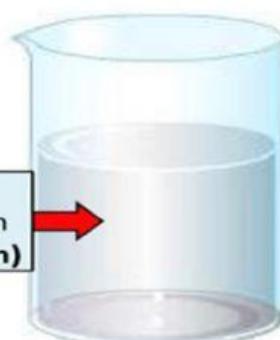
Water (solvent)

Sugar (solute)

All other substances in a solution

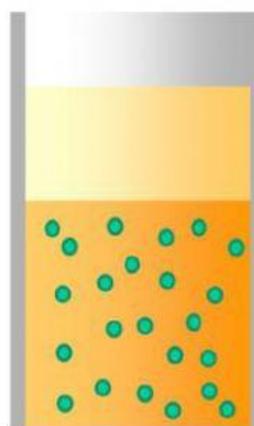


Sugar solution (solution)

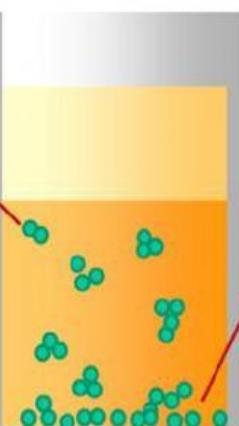


Example of a stable colloid

suspension



Aggregation



Sedimentation

## Drag and drop

Heterogenous

Homogenous

Suspension

Colloid



### Mixture

A mixture that is the same everywhere



### Solution

### Mixture

A mixture is **NOT** the same everywhere



The mixture is cloudy.



You can see the parts of the mixture clearly.

