

1 Lengkapkan ruang tentang struktur suatu atom neutral. TP1

Complete the space about the structure of a neutral atom.



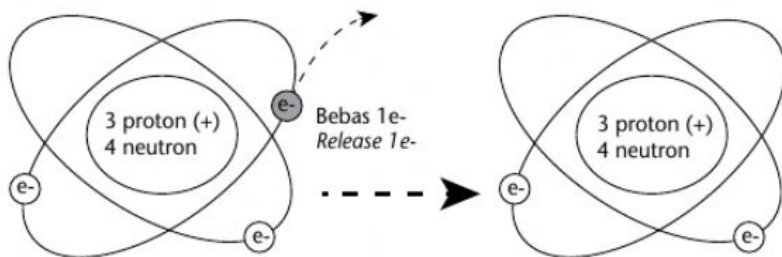
Video

Positif Positively	Neutron Neutron	Elektron Electron	Proton Proton	Negatif Negatively
(a) _____ (Zarah yang bercas charged particle)				(b) _____ (Zarah yang bercas charged particle)
Bilangan elektron = bilangan proton Number of electrons = number of protons				(c) _____ (Zarah yang tidak bercas. Particle with no charge)

Nukleus
Nucleus

2 Lengkapkan ruang tentang pembentukan ion positif (kation). TP2

Complete the spaces about the formation of positively charged ion (cation).



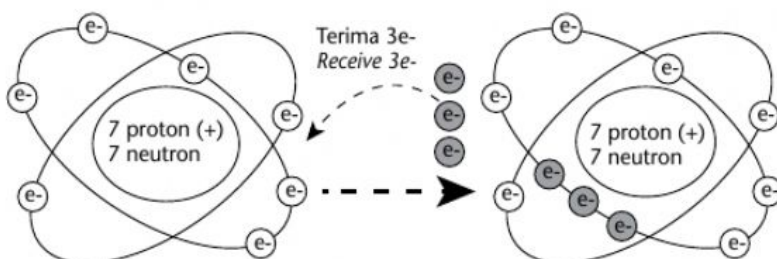
Atom neutral, 3p (+) = 3e (-)
Neutral atom

Cas bagi 3 proton =
Charge of 3 protons
Cas bagi 2 elektron =
Charge of 2 electrons
Jumlah cas =
Total charge

- Lebihan ____ cas positif menyebabkan ____ terbentuk.
Excess of ____ positive charge causes a ____ to form.

3 Lengkapkan ruang tentang pembentukan ion negatif (anion). TP2

Complete the spaces about the formation of negatively charged ion (anion).



Atom neutral, 7p (+) = 7e (-)
Neutral atom

Cas bagi 7 proton =
Charge of 7 protons
Cas bagi 10 elektron =
Charge of 10 electrons
Jumlah cas =
Total charge

- Lebihan ____ cas negatif menyebabkan ____ terbentuk.
Excess of ____ negative charges causes a ____ to form.



Praktis
Kendiri