Chapter 6: periodic table.

Classification of elements.

1. complete the statements below.

Matter (solid, liquid, gas)	Consists of small and discrete particles called as
Atom is consis	sts of three subatomic particles which are
(a)	(b)(c)
	Characteristic of subatomic particles:
	(a) electron :
	(b) proton:
	(c) neutron :
(c)	(d)
	are found in the nucleus of an atom.
(b)surrou	
	charges due to the positively-charged proton in it.
(a) Number of proton = number of electro	on. Hence, an atom is
are neutral particl	es made up oforatoms.

Metals and non-metals

1. Compare the characteristic of metal and non-metal

Metals	Similarity	Non-metals
Metals	Characteristics	Non-metals
	Surface properties	
	Ductility	
	Malleability	
	Tensile strength	
	Electrical conductivity	
	Heat conductivity	
	Melting and boiling point	
	Density	

6.2. Mixture

1. Mixture consists	or	elements or
compounds mixed.		

2	As the mixture is	formed physically.	t can be separated	too.

METHOD TO SEPARATE













